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Saline Inflows, Biodegradation and Gas Generation in the Basal Water Sand Aquifer, Athabasca Oil Sands Region, Alberta



Outline

- Introduction
- Geology
- Hydrogeology
- Geochemistry
- Industrial Significance
 - Biodegradation
 - Gas Generation
 - Saline Inflows
- Summary



[Archives, Government of Canada](#)



Key Messages – Basal Water Sand (BWS)

Geology

- Deposited on a highly dissected erosion surface cut into a limestone karst terrain; distribution influenced by both pre- and post-depositional structure

Hydrogeology

- Regional recharge of the BWS east and west of the Athabasca River
- Regional discharge along the Athabasca River where the BWS experiences upward flow of basinal Devonian brines

Geochemistry

- The BWS occurs within the interface between:
 - A hypersaline, oxidized Paleozoic geochemical network, and
 - A fresh to brackish, reducing Mesozoic geochemical network
- In the discharge region, secondary minerals record vertical movement and progressive reduction of oxidized saline brines



Significance to Industry

Oil sands mining

- Saline inflows
- BWS depressuring
- On-site water storage / transfer

In situ

- Brackish groundwater resource
- Potential thief zone

Geochemical Processes

- A locus for water-rock-microbial interactions:
 - Reduced bitumen saturations
 - Biodegradation in discharge (upflow) areas
 - H₂S generation associated with biodegradation



Basal Water Sand

Oil Sand



Basal Water Sand





Location

- Basal Water Sand (BWS) aquifer located in the surface mineable oil sands area
- The BWS may or may not be present in the Athabasca south region and other areas

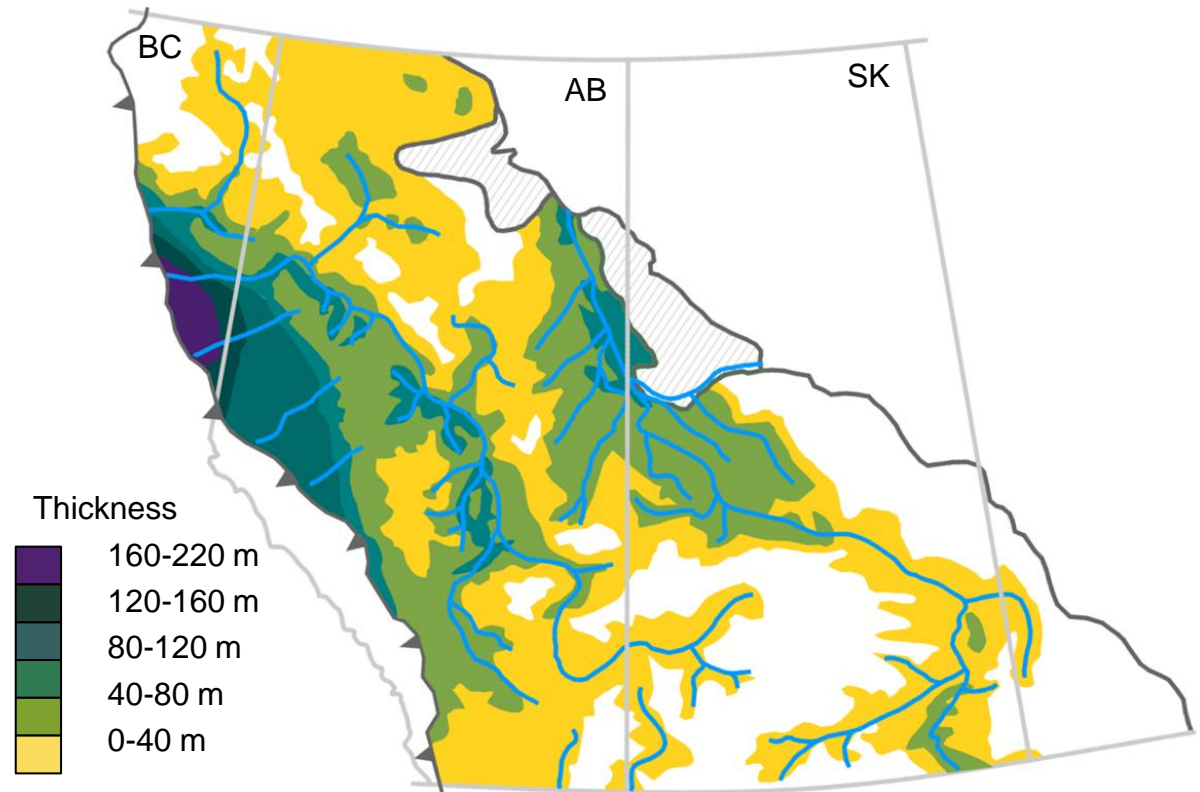


Government of Alberta



Lower Mannville Isopach

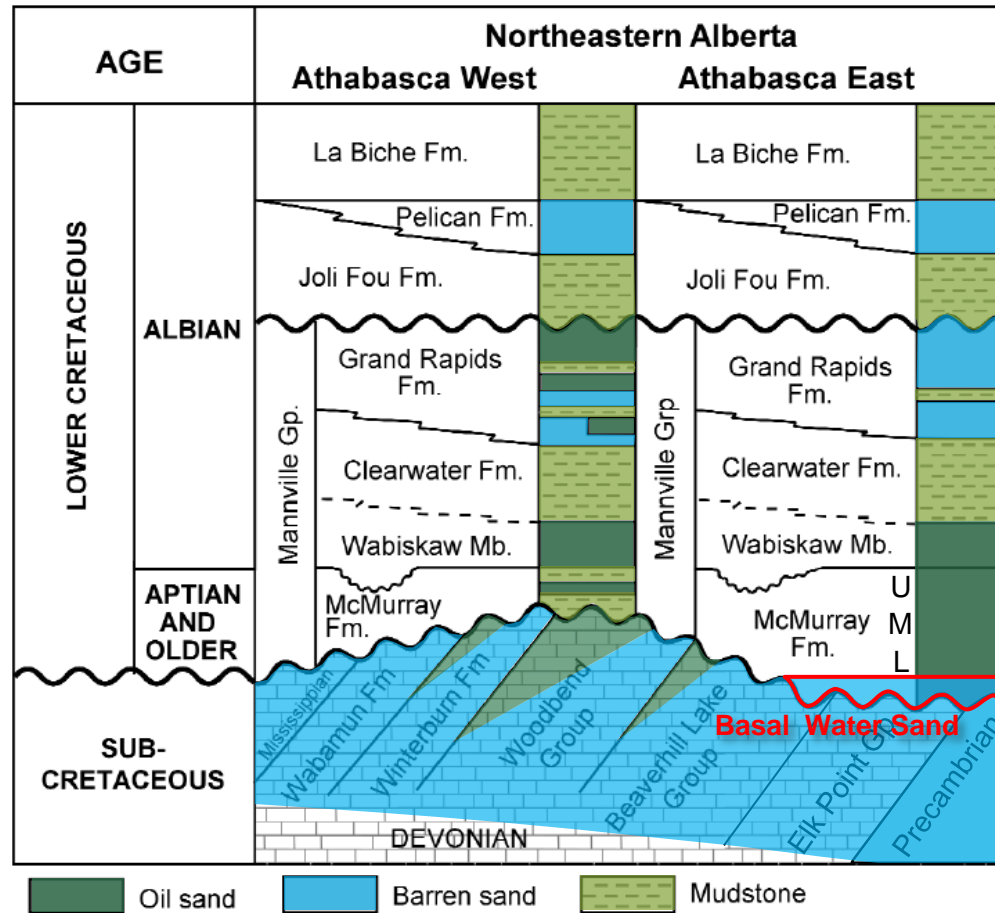
- LMnv deposition controlled by salt dissolution (east) and flexural subsidence (west)
- LMnv in NE Alberta includes Basal Water Sands deposited in N prograding delta
- LMnv sediments predominantly supermature quartz derived from the Precambrian Shield



After Cant and Abrahamson (1996)



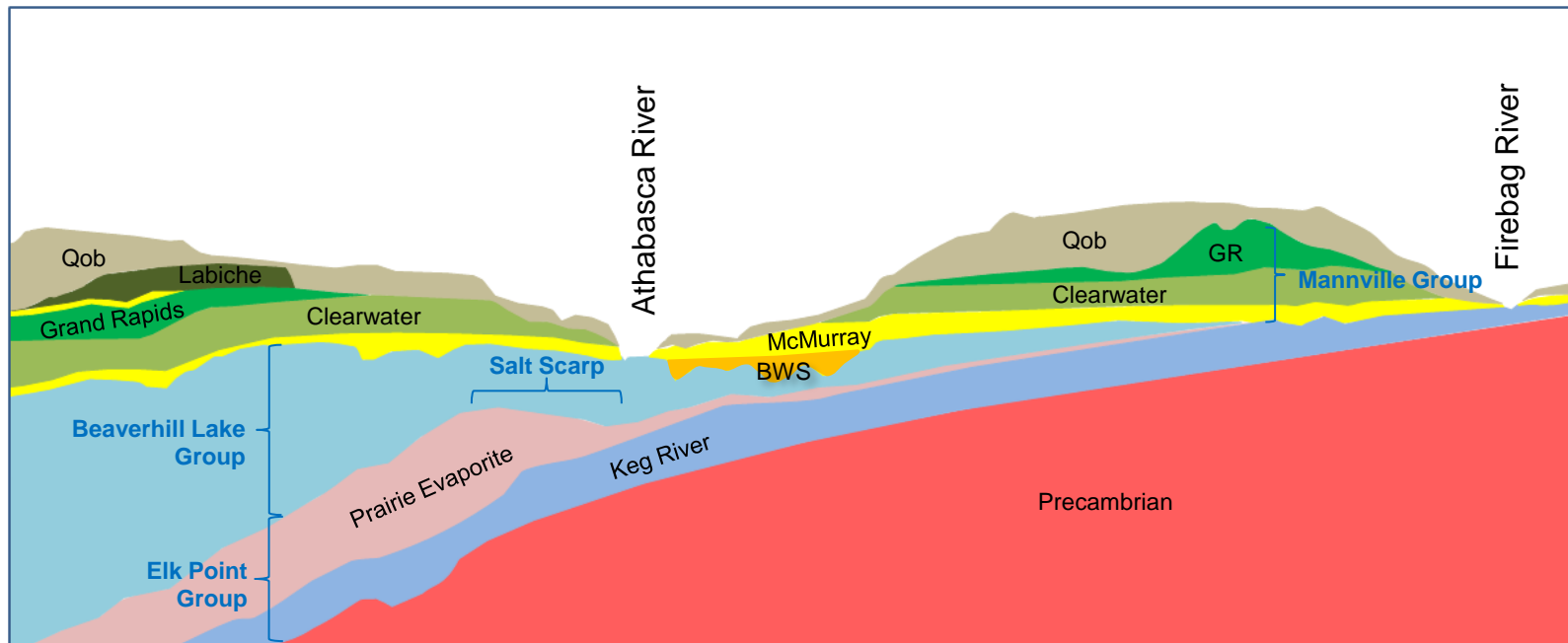
Stratigraphic Column, Northeast Alberta



After EUB/AGS Earth Sciences Report 2001-06



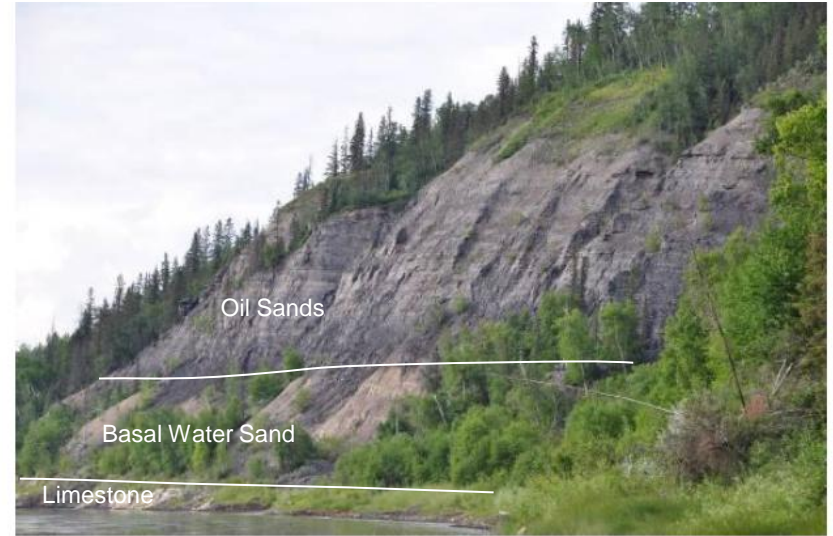
Schematic Geological Cross Section



- Simplified geological cross section north of Fort McMurray
- BWS aquifer sits unconformably on structured Devonian limestone



Basal Water Sand in Outcrop

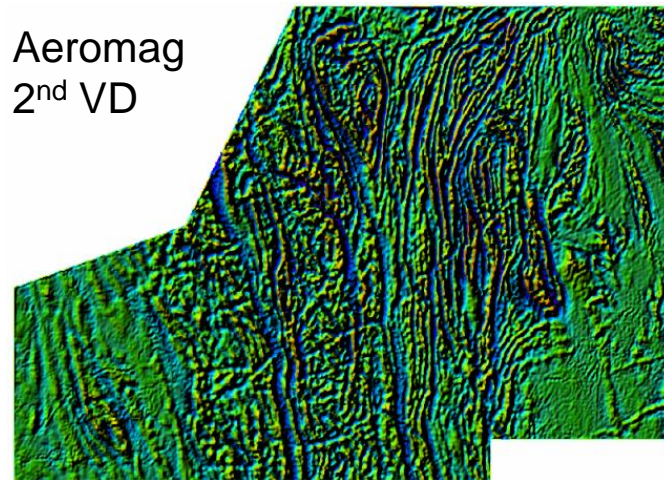


Gibson et al., OSRIN (2011)

- Cretaceous McMurray Formation of the Mannville Group exposed along the bank of the Athabasca River
- The McMurray Formation sits unconformably above Devonian limestone of the Waterways Formation, Beaverhill Lake Group



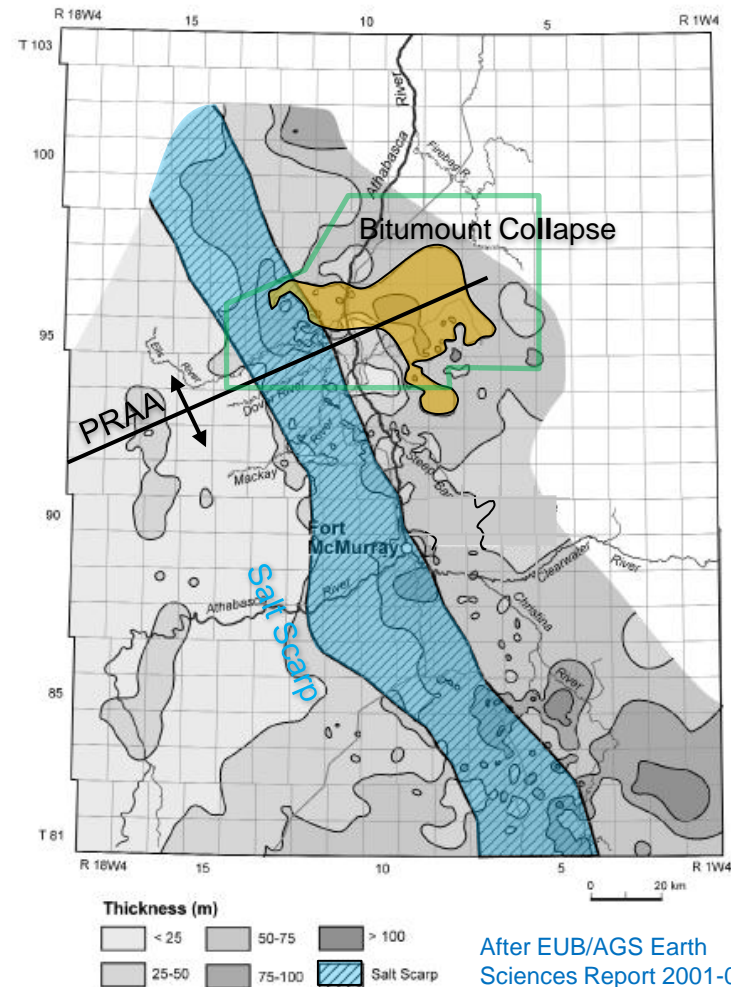
Regional Structural Elements



Aeromag
2nd VD

Birch Mountain Resources
(1997)

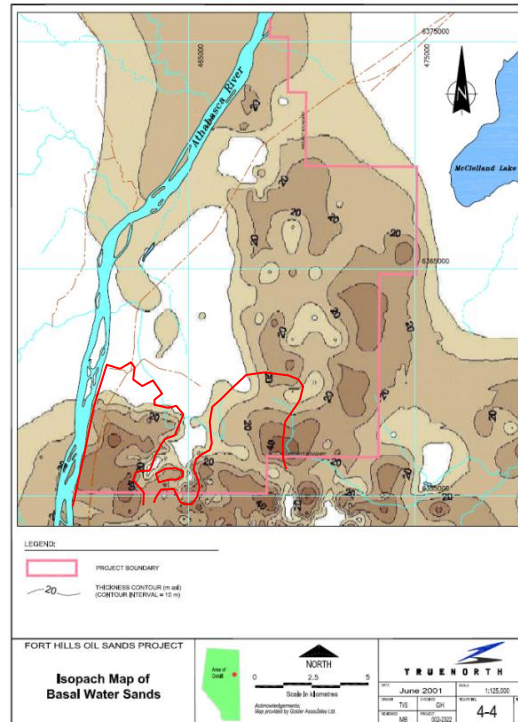
- Abundant evidence for structural control of BWS distribution
- Salt solution collapse
- Potential basement involvement



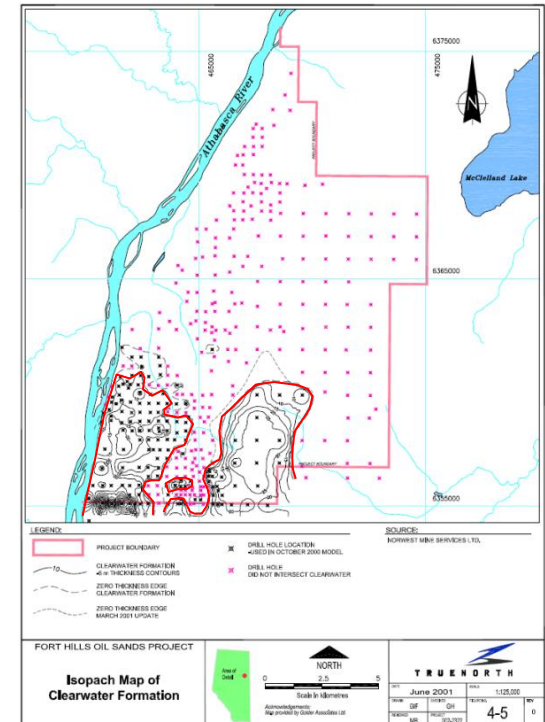
Structural Control of Basal Water Sand

- Clearwater Formation isopach appears to template thickest BWS at Fort Hills
- Clearwater Formation is preserved as an inlier
- Structural down-dropping of Clearwater may have led to thicker BWS development

Basal Water Sand



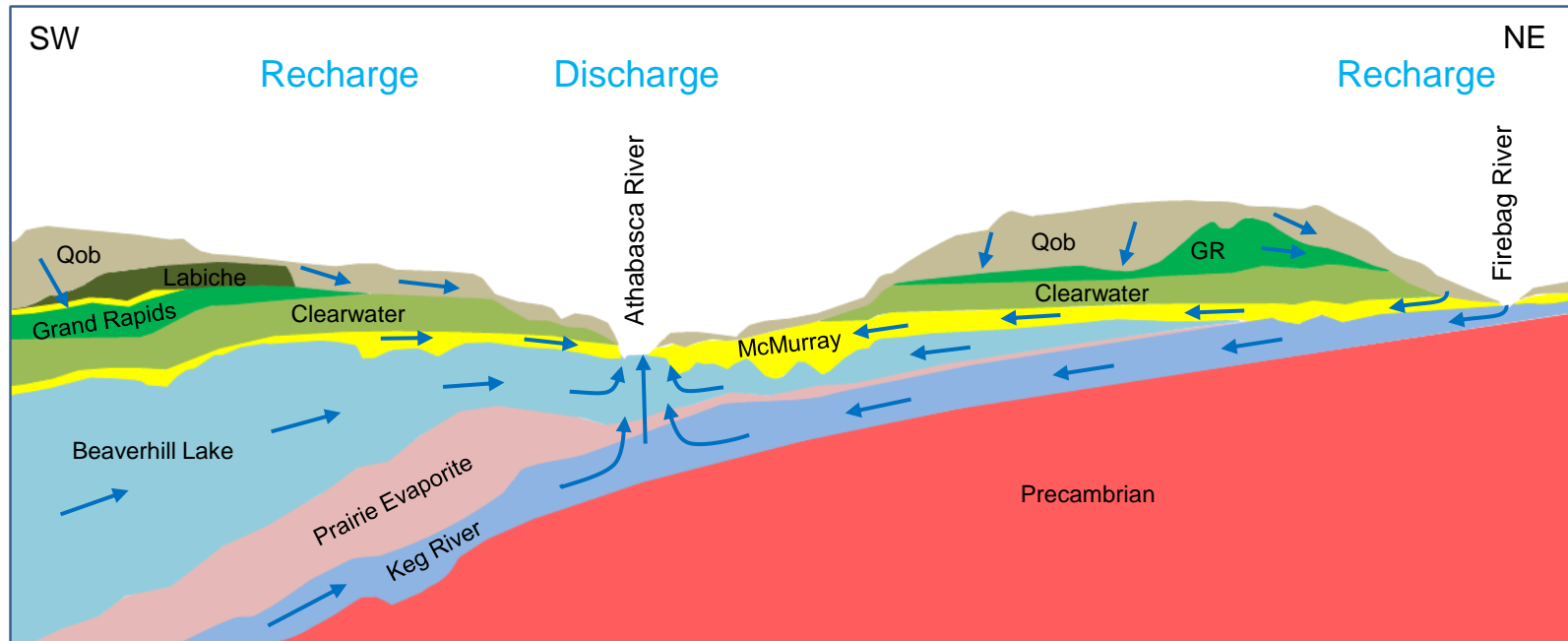
Clearwater Formation



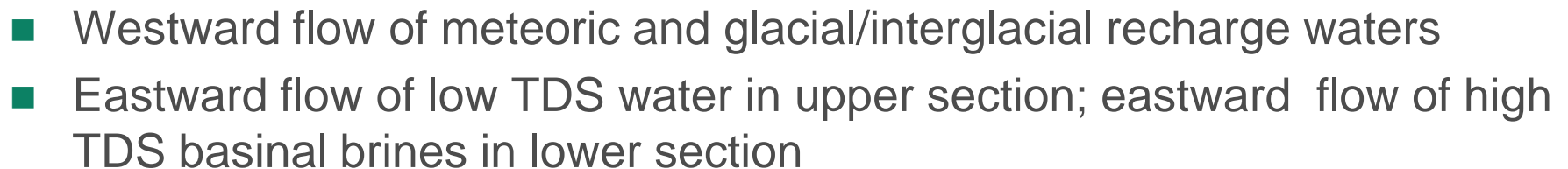
Deer Creek Energy (2001)



Regional Hydrogeological Setting



- Two major flow systems discharge to the Athabasca River
 - Deep basin flow system (Devonian)
 - Shallow surface flow system (Cretaceous to Quaternary)



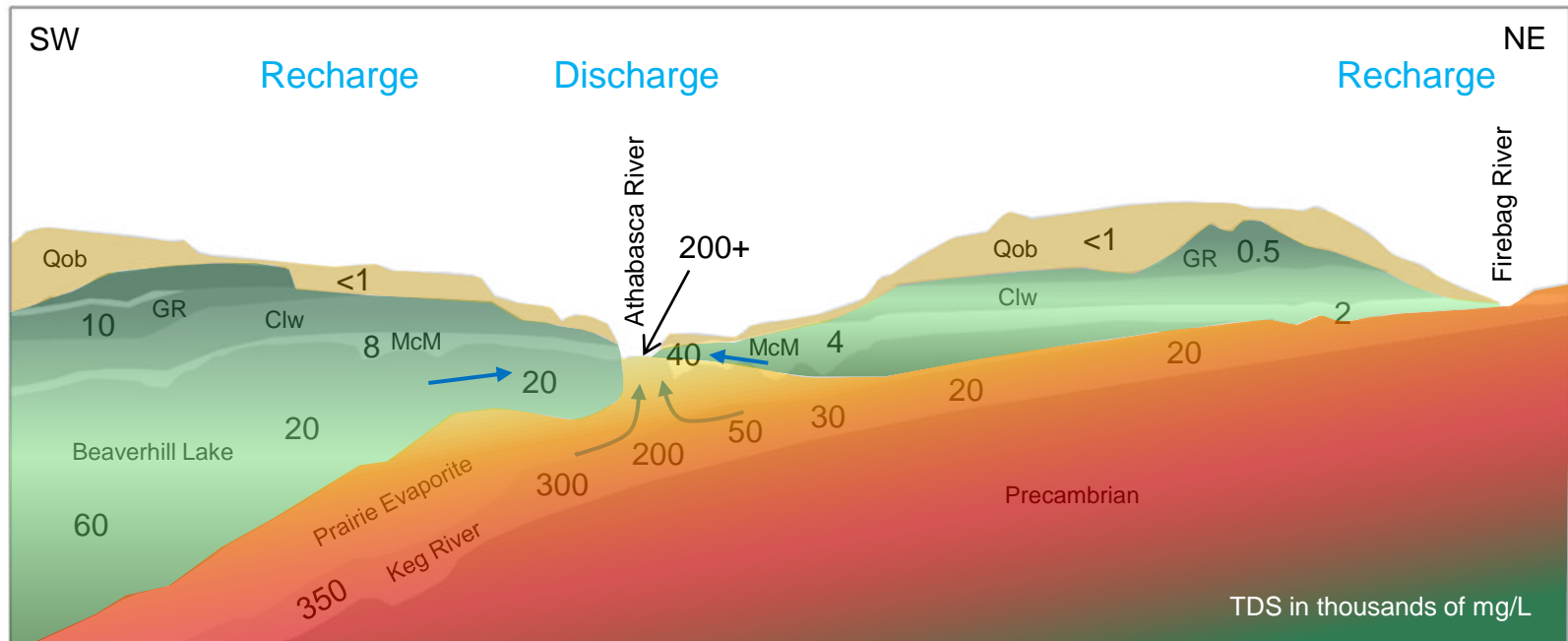


Brine Discharge at La Saline Springs





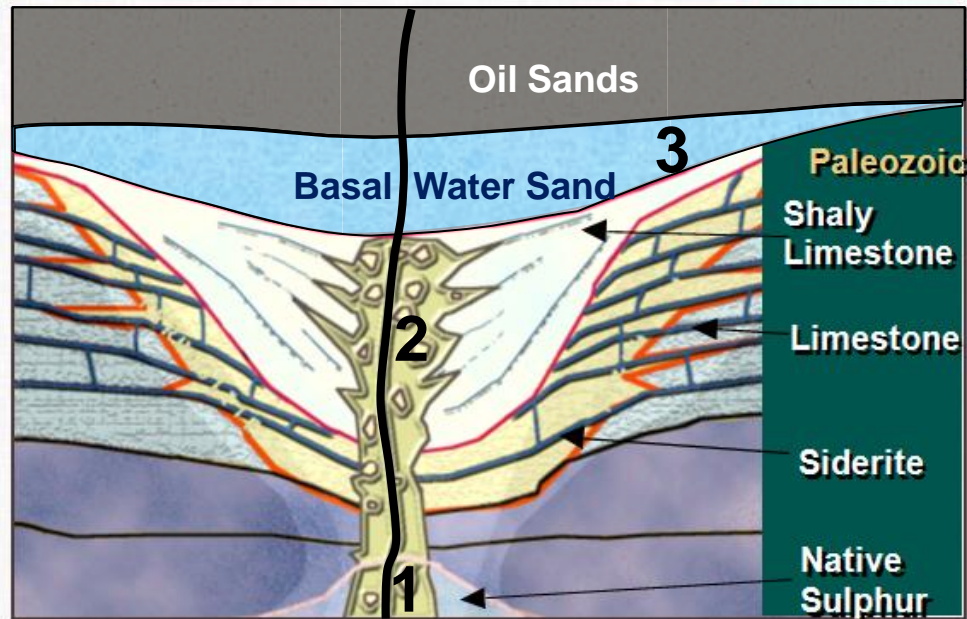
Simplified Redox Stratigraphy



- Oxidized Elk Point brines (red) and surface waters
- Reduced conditions (green) in rocks in contact with bitumen and deep in basement



Redox-Driven Mineral Alteration



- Vertical flow of oxidizing Na-Ca-Cl-SO₄ brines:
 - Oxidation of primary pyrite,
 - In the presence of bitumen, bacterial sulphate reduction produces H₂S and CO₂
 - H₂S reacts forming secondary pyrite and pyrrhotite,
 - CO₂ reacts to form H⁺ and HCO₃⁻, causing decalcification and siderite precipitation

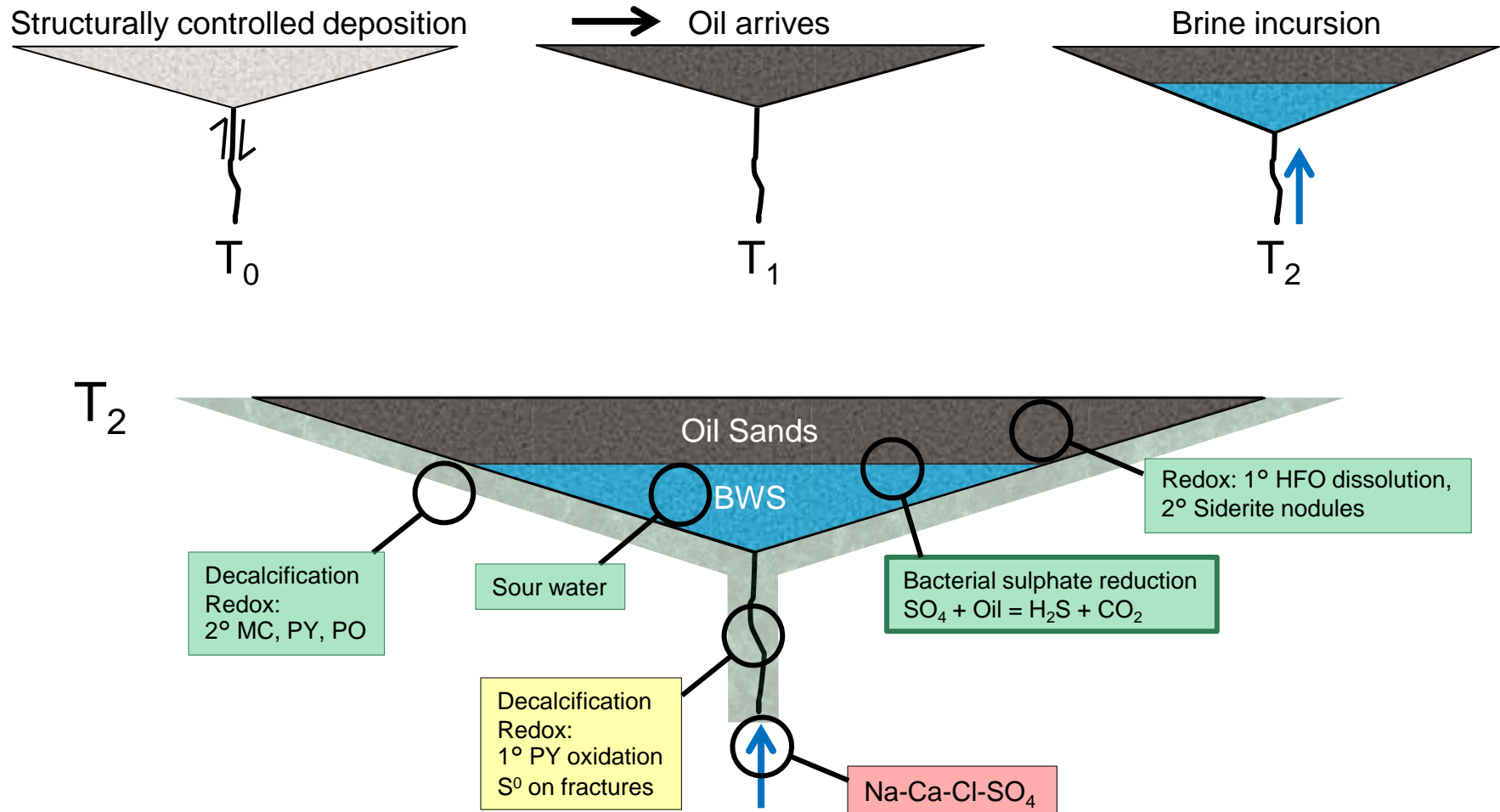


SECONDARY MINERALS

1. Native S, Keg River Formation.
2. Pyrrhotite, Waterways Formation.
3. Siderite nodules, Lower McMurray Formation.



BWS Conceptual Geochemical Model





Industrial Impact of BWS

Oil sands mining

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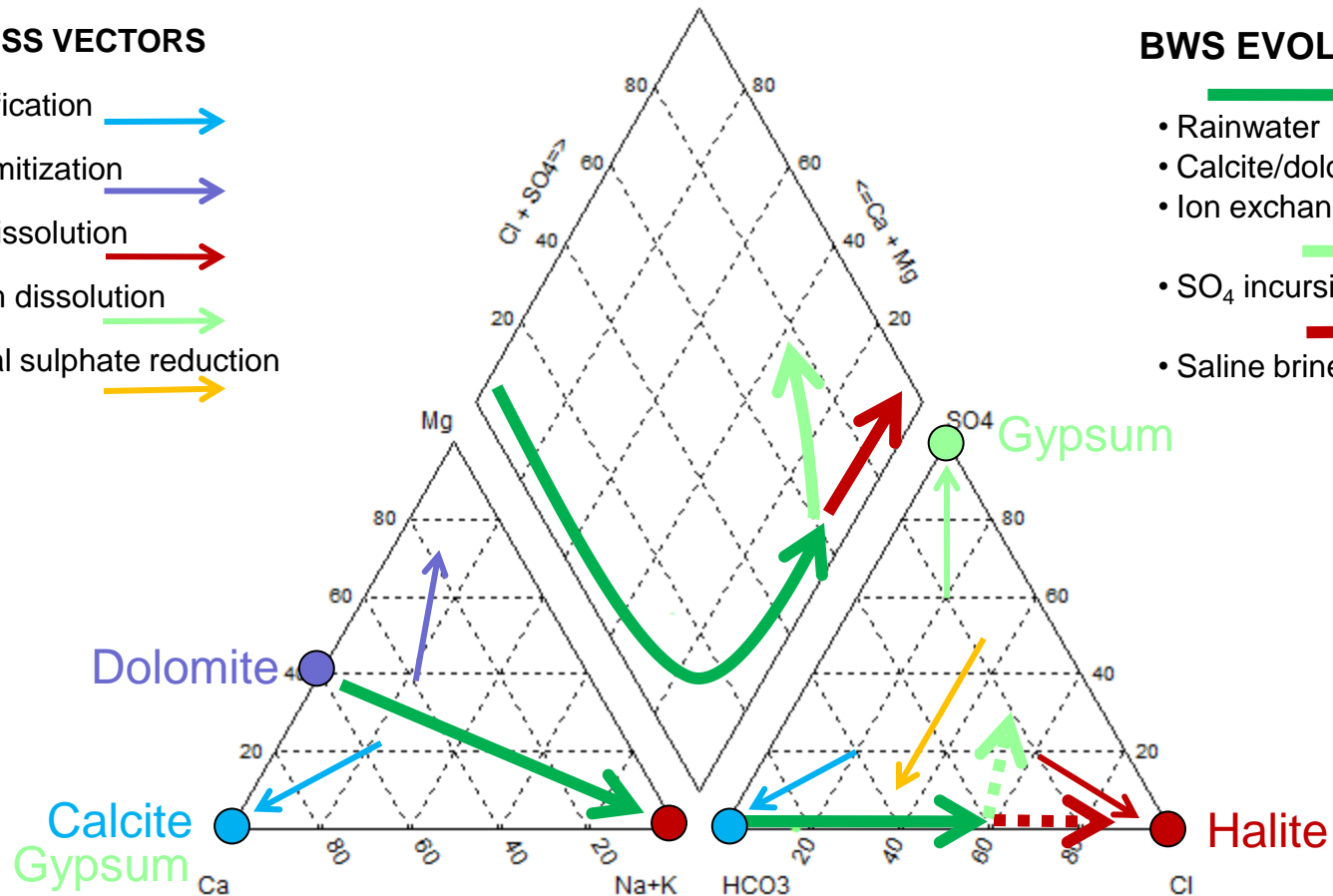


Archives, Government of Alberta



- Decalcification
- Dedolomitization
- Halite dissolution
- Gypsum dissolution
- Bacterial sulphate reduction

- Rainwater
- Calcite/dolomite
- Ion exchange
- SO₄ incursion
- Saline brine incursion



Piper diagram - mass units



- ## BWS EVOLUTION

- Rainwater
- Calcite/dolomite
- Ion exchange
- SO₄ incursion
- Saline brine incursion





Modeling Bacterial Gas Generation

- PHREEQC used to model gas bubble generation via reaction with bitumen
- Henry's Law K_H values used for CH_4 , CO_2 , H_2O , H_2S and N_2
- Incremental addition of bitumen to reach supersaturation with the least soluble gas (CH_4)
- Track total mass and volume of gaseous species in the gas phase
- Decrease containing pressure to simulate depressuring

Reaction 1. Irreversible reaction defined in simulation 4.

3.000e-002 moles of the following reaction have been added:

Reactant Relative 30 millimoles of...
 moles
(C) 2 (H) 2.93 (N) 0.01 (S) 0.04 (O) 0.02 1.00000 ...model bitumen...

Element Relative
 moles
C 2.00000
H 2.93000 ...titrated into 1 Kg of
N 0.01000 BWS water...
O 0.02000
S 0.04000

-----Gas phase-----

Total pressure: 15.0000 atmospheres ...at 15 bar pressure...

Gas volume: 2.54e-002 liters

...results in initial bubble formation...

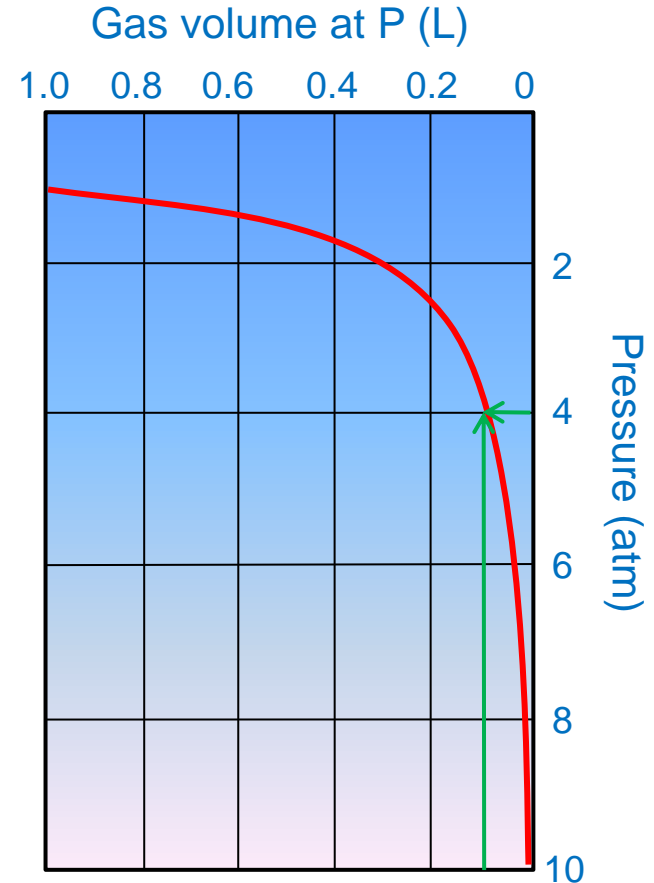
 Moles in gas

Component log P P Initial Final Delta
...of CH_4 with minor CO_2 .
CH4 (g) 1.13 1.364e+001 0.000e+000 1.489e-002 1.489e-002
CO2 (g) 0.12 1.332e+000 0.000e+000 1.454e-003 1.454e-003
H2O (g) -1.92 1.189e-002 0.000e+000 1.299e-005 1.299e-005
H2S (g) -1.69 2.062e-002 0.000e+000 2.251e-005 2.251e-005
N2 (g) -5.05 8.946e-006 0.000e+000 9.768e-009 9.768e-009



Depressuring Model Results

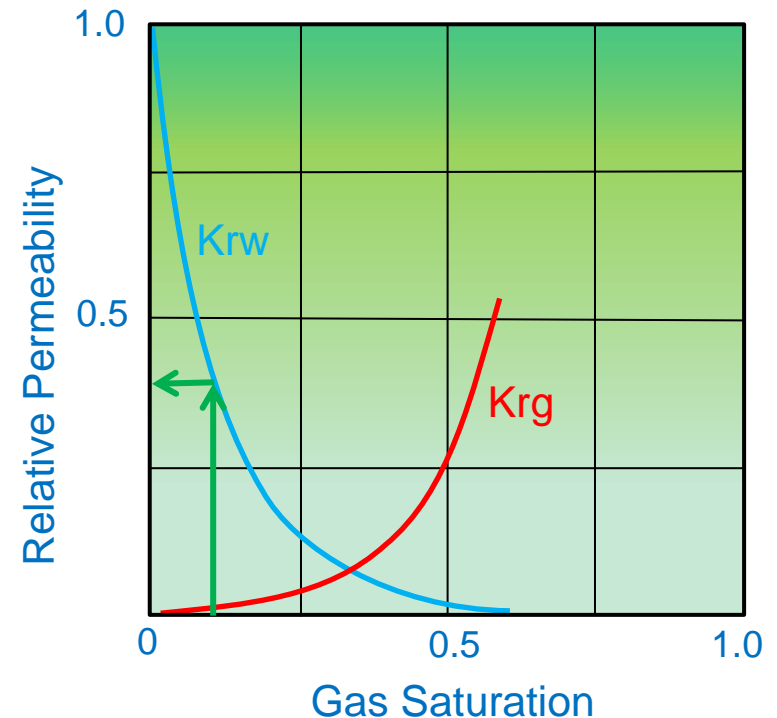
- Gas volume at saturation calculated for ~90 m depth (10 atm)
- Gas volume tracked as pressure is reduced; i.e., drawdown
- At ~60 m drawdown gas volume is ~10% of total volume
- This analysis does not consider the salting-out effect which further decreases gas solubilities





Implications for Depressuring

- Biodegradation of small quantities of bitumen may provide sufficient gas to saturate BWS at depth
- Schematically, a pumping drawdown of 60 m on a gas-saturated BWS water at 90 m depth can result a loss of 60% relative permeability to water
- Analysis is qualitative, but illustrates the importance of understanding BWS, including biodegradation, gas generation and salinity



P. Glover, Formation Evaluation
M.Sc. Course Notes.



Saline Inflows – The Devonian Connection

- Operators in the Mineable Oil Sands area have, from time to time, encountered inflows of saline water inferred to originate in Devonian aquifers
- BWS provides part of the pathway linking Pz and Mz formations
- Present-day salinity distribution suggest little connection between Devonian and McMurray Formations
- Mineralogy, bitumen chemistry and structure imply in the past there may have been connection
- Prior structural / hydrogeological fairways may be reactivated during mining – multiple layered datasets are needed to locate these fairways



BWS Summary

- Regionally, the BWS is subjected to numerous physical, chemical and biological processes:
 - Meteoric recharge to the east and in elevated areas to the west
 - Interglacial to early post-glacial recharge
 - Upflow of Devonian brines, themselves influenced by
 - Decalcification and dedolomitization
 - Gypsum and/or halite dissolution
 - Bacterial sulphate reduction is expected where upflowing waters carrying sulphate encounter bitumen, leading to:
 - Production of H_2S and CO_2
 - Reductive dissolution of hydrous ferric oxide minerals
 - Precipitation of sulphide minerals
- Site-specific knowledge of Devonian and BWS waters is required to address key issues of concern to oil sands industry, regulators



Questions...



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