

WATERtech 2012  
Banff, AB



# **Dissolved Organic Matter Characterization of SAGD Produced Water**

## **Effects on Produced Water Treatment**

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# Outline

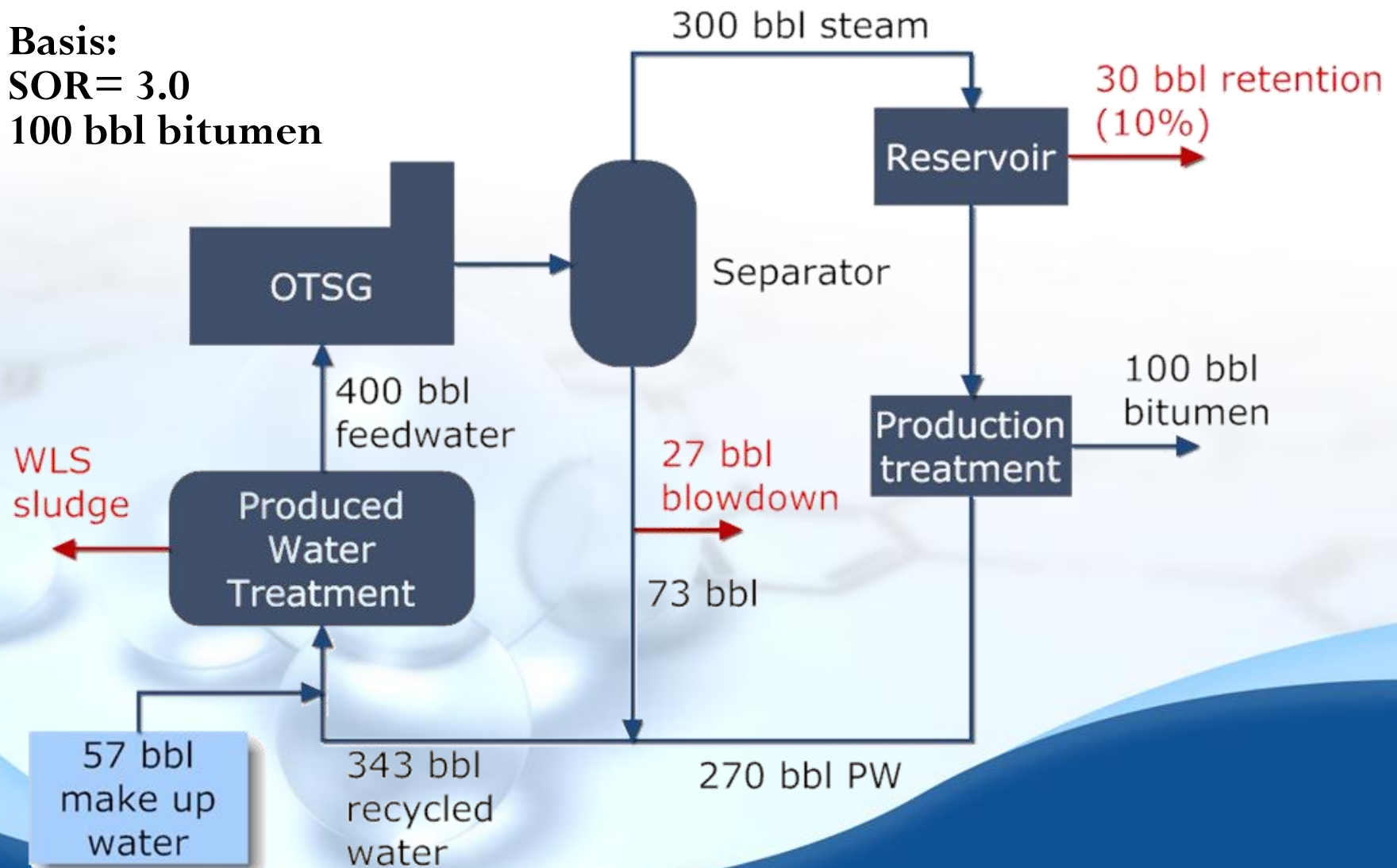
- ◆ Overview of SAGD produced water (PW) treatment
- ◆ Water chemistry management challenges:
  - Reduce blowdown water disposal
  - Reduce make up water
  - Improve reliability of treatment processes
- ◆ Knowledge of water chemistry
  - Snapshot of the chemistry of blow-down water
- ◆ Dissolved organic matter (DOM)
- ◆ Concluding remarks

# Typical SAGD Surface Treatment

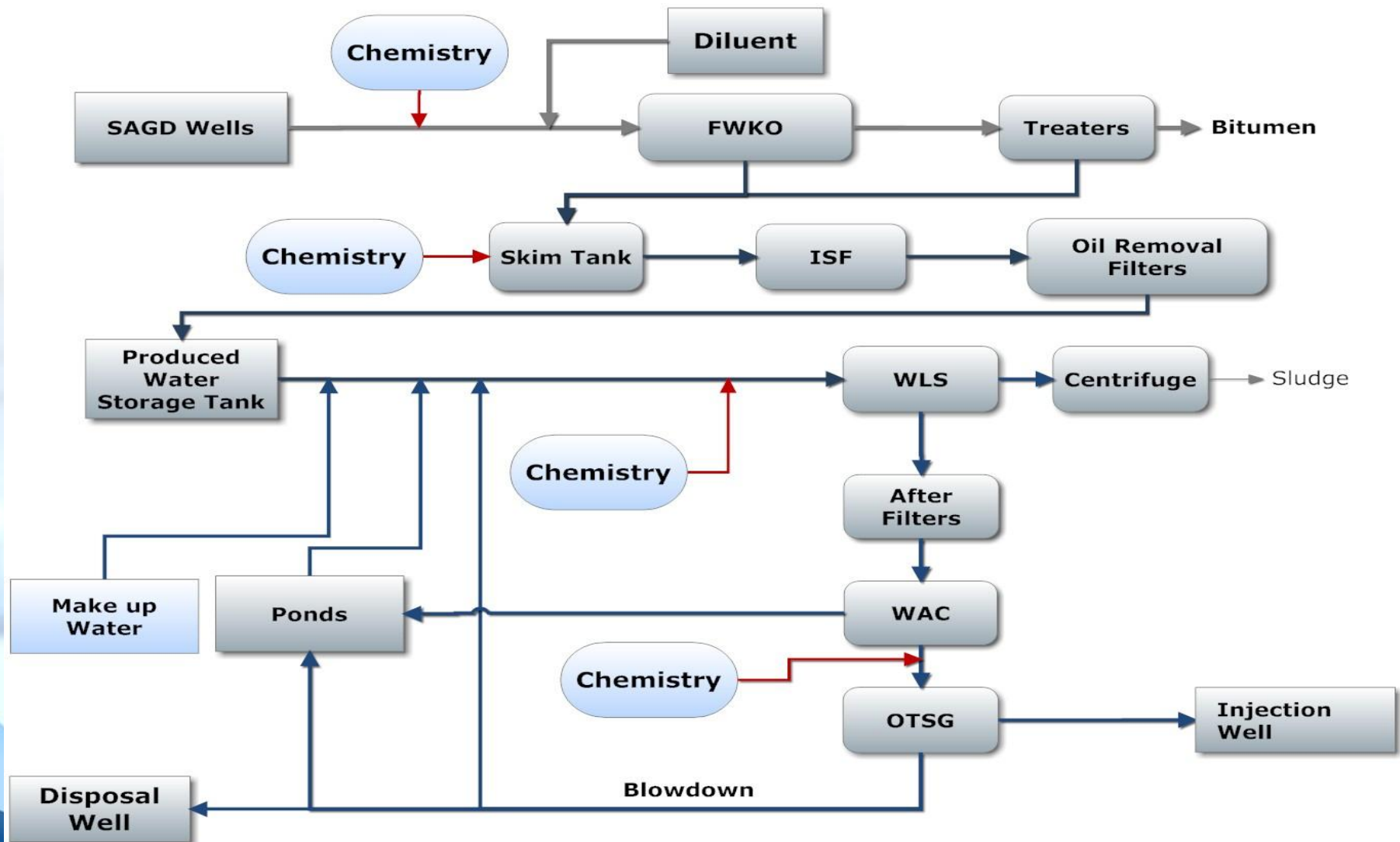
Basis:

SOR= 3.0

100 bbl bitumen



# Schematic of Surface Treatment



# Characteristics of Different Oil Field Produced Water and SAGD Blow-Down

Characteristics	Oil Field Produced Water <sup>1</sup>	Oil Sands Produced Water <sup>2</sup>	SAGD BBD (Our Study)
pH	7.4-8.5	7.11	11.2-10.3
Conductivity ( $\mu\text{S}/\text{cm}$ )	1400-5000	1540	10000-18000
TSS (mg/L)	97	-	25-65
TDS (mg/L)	700-2000	-	12000-17000
TOC (mg/L)	68-140	232	500-2500
Silica (mg/L as Si)	7-14	-	94-256

<sup>1</sup>Mondal, S., Wickramashingh, S. (2008) J. Membr. Sci. 322, 162-170

<sup>2</sup>Petersen, M.A., Grade, H. (2011) Ind. Eng. Chem. Res. 50, 12217-12224

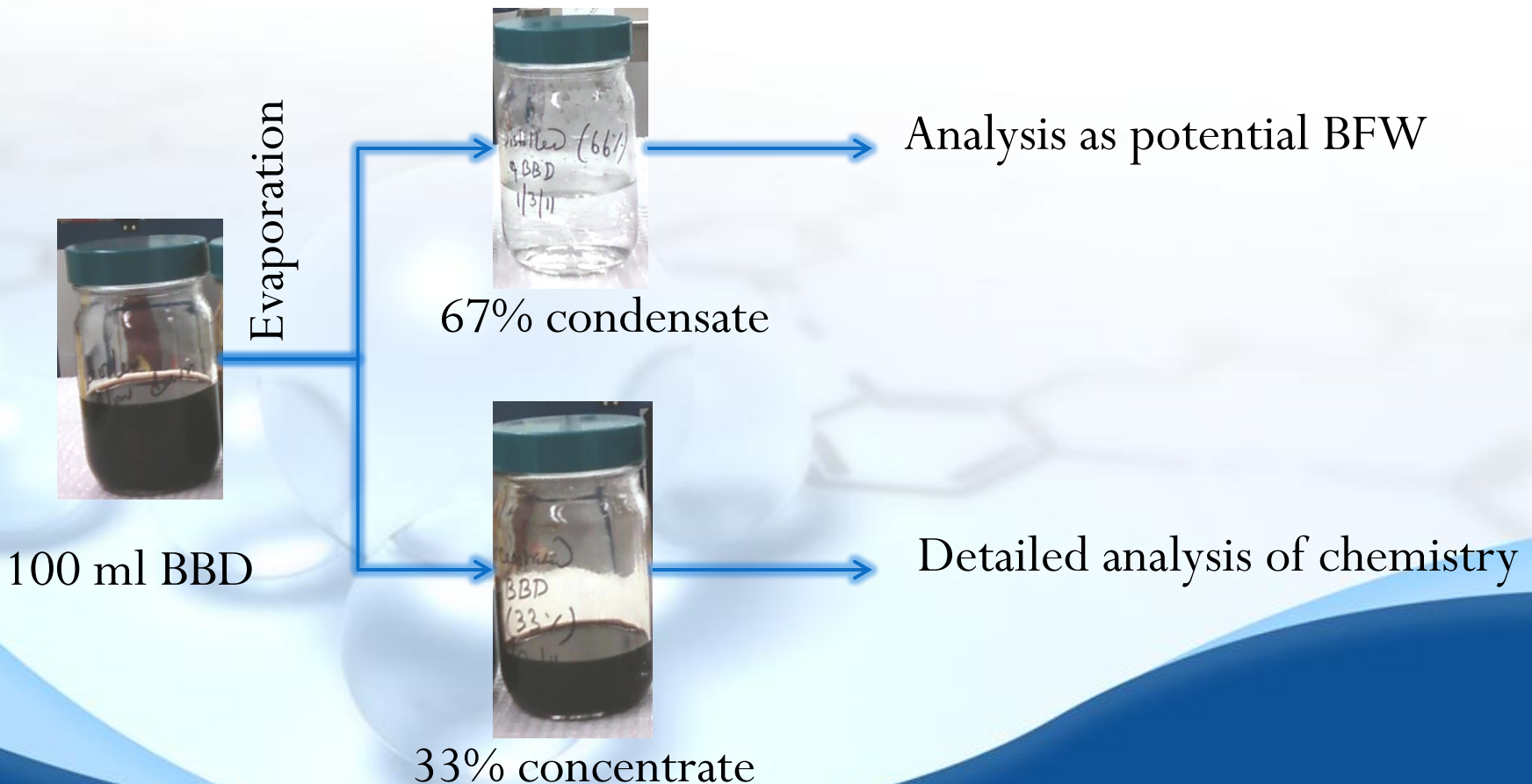
# **UNDERSTANDING THE CHEMISTRY OF SAGD PW**

Chemistry of the Blow Down Water

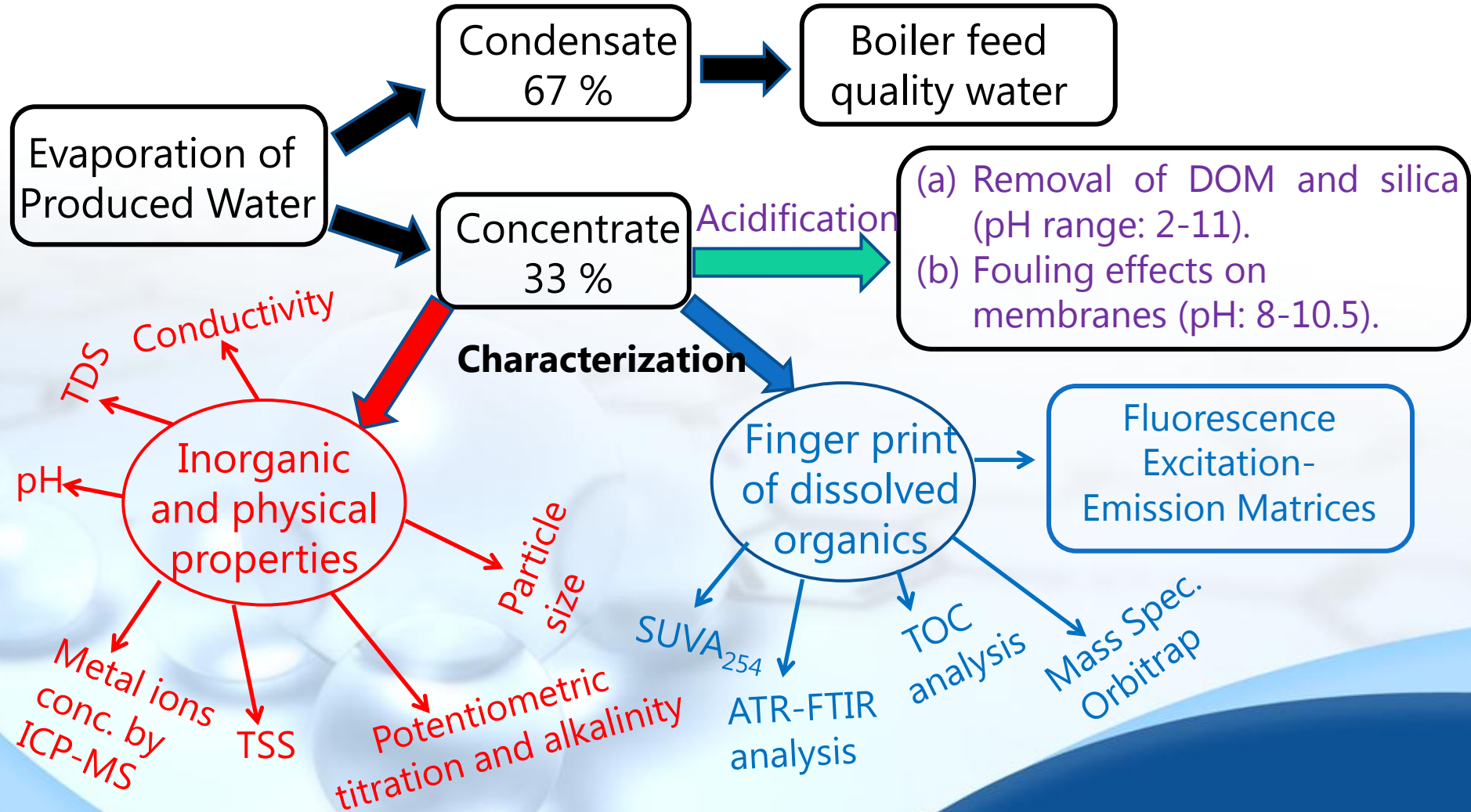


# SAGD Blowdown Water Evaporation

**“End of the pipe” water – represents cumulative effect of all chemistries.**



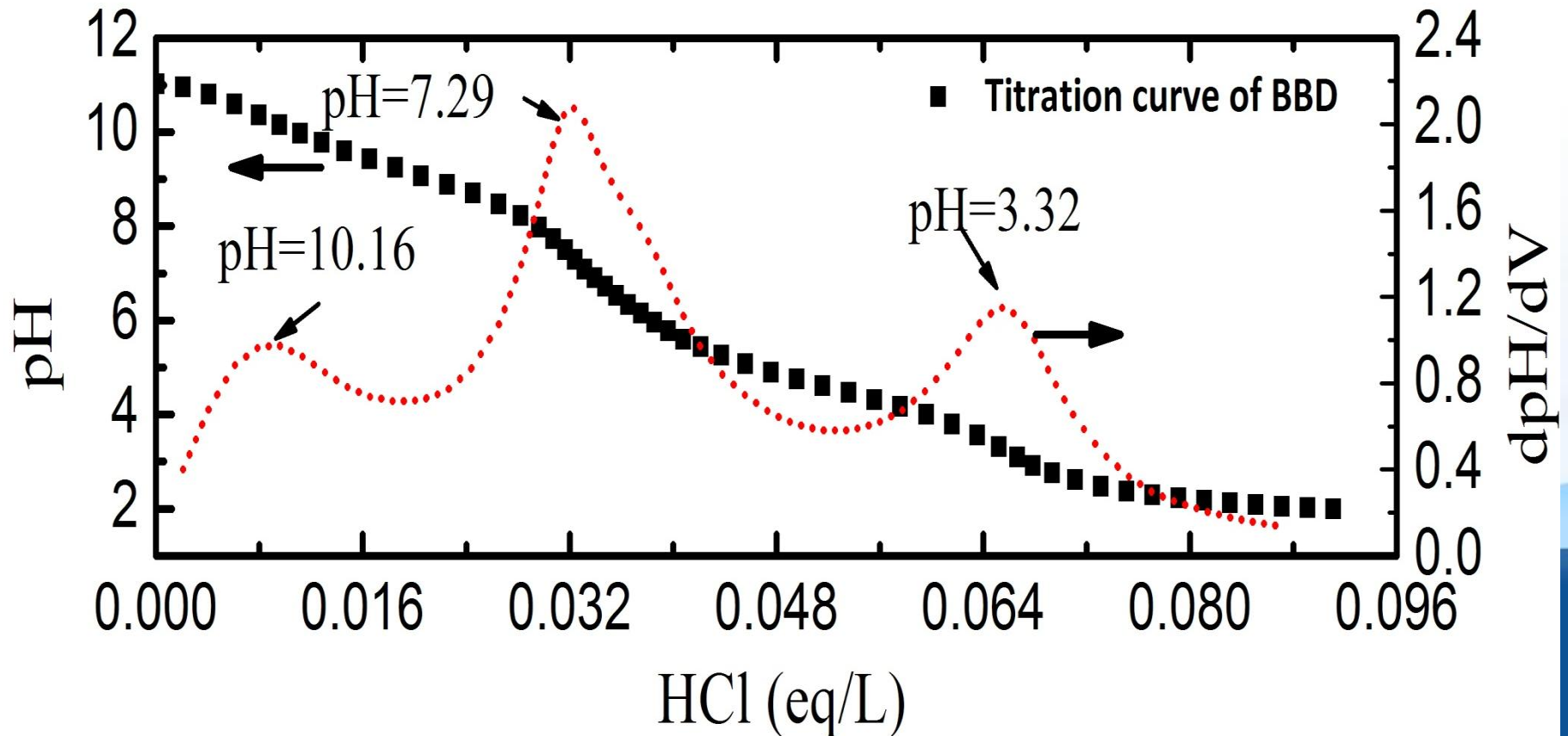
# Characterization Methods



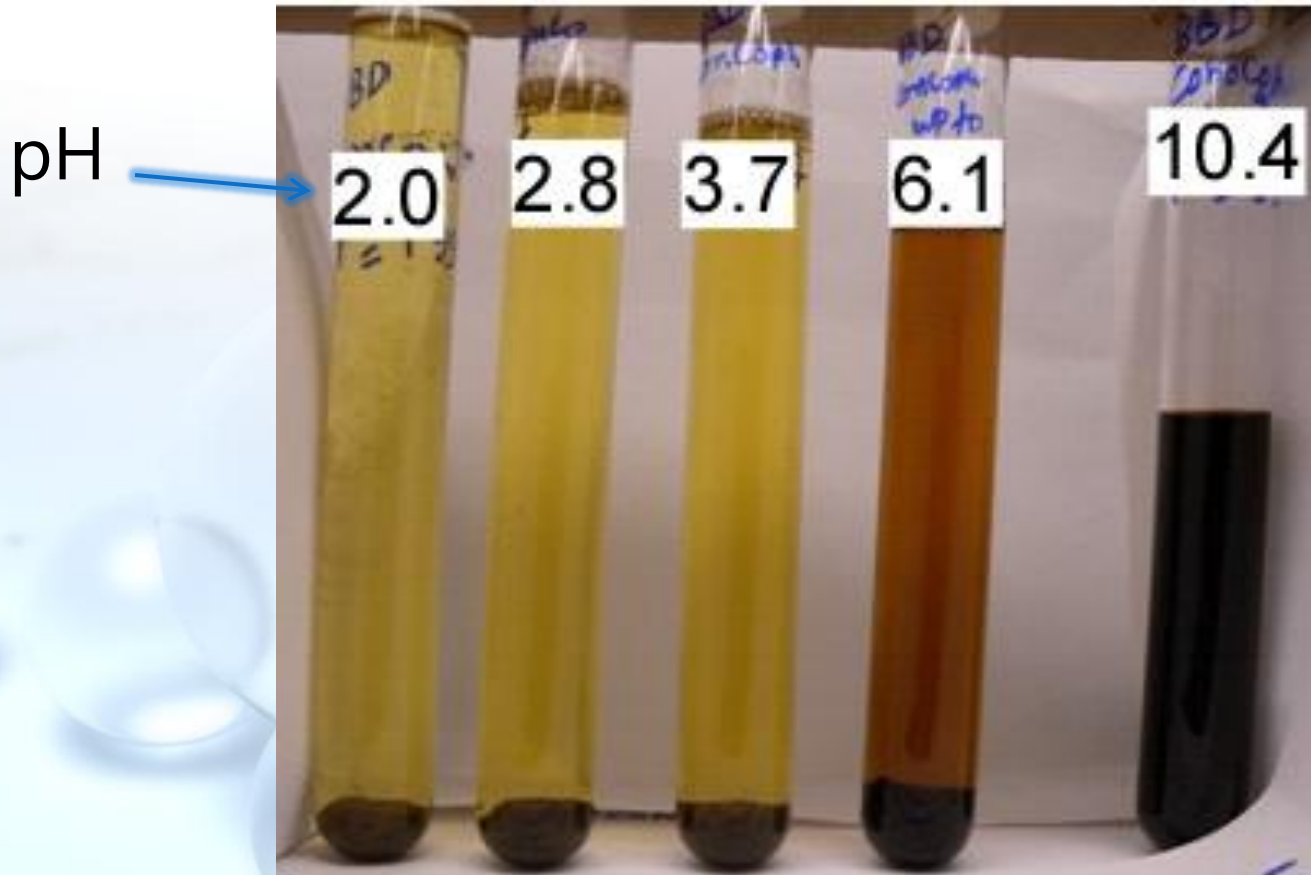


# Potentiometric Titration of BBD

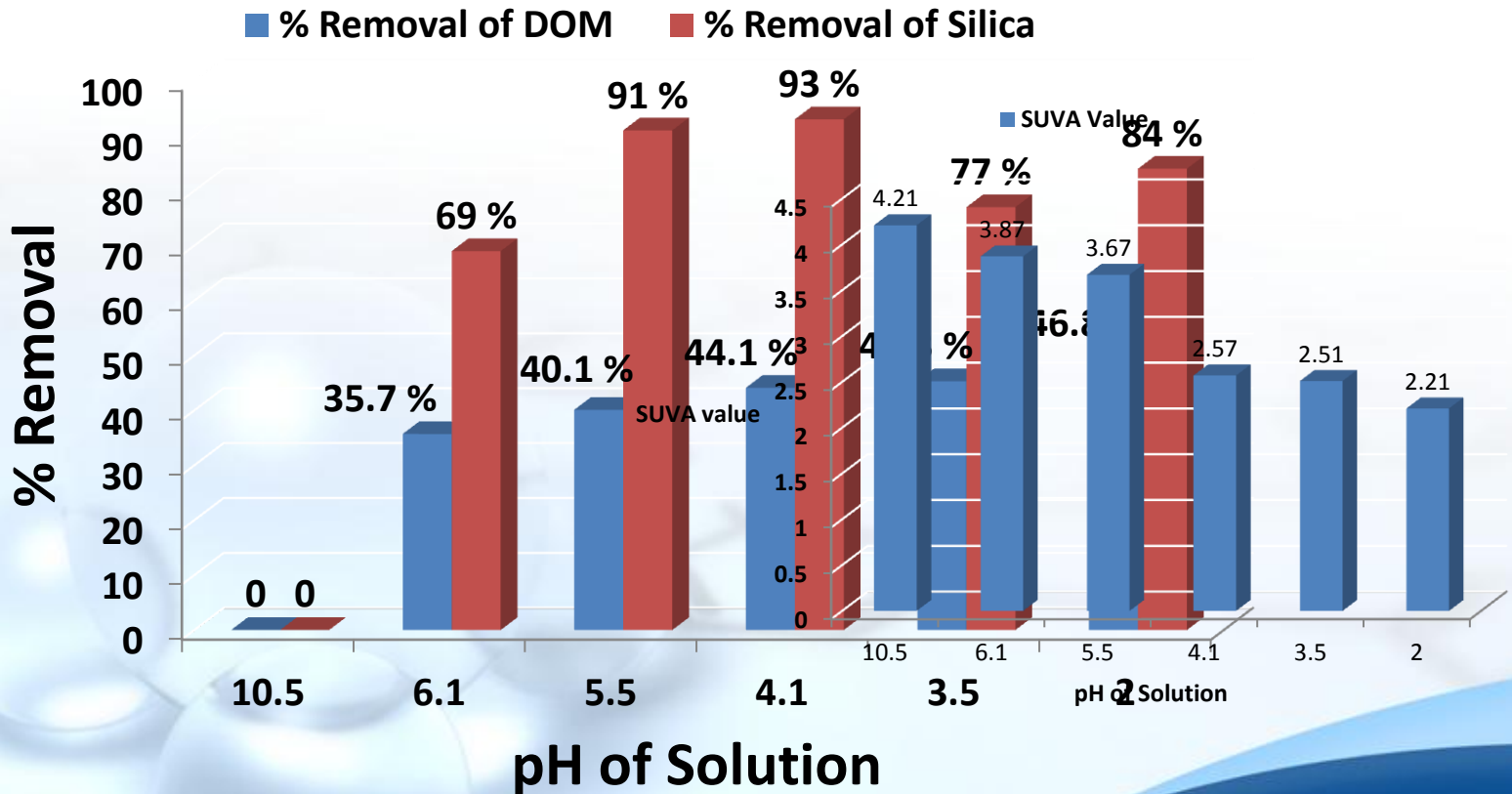
Acid-titration of BBD water performed to different end-points using a potentiometric "true equilibrium" auto-titrator (QC-Titrator)



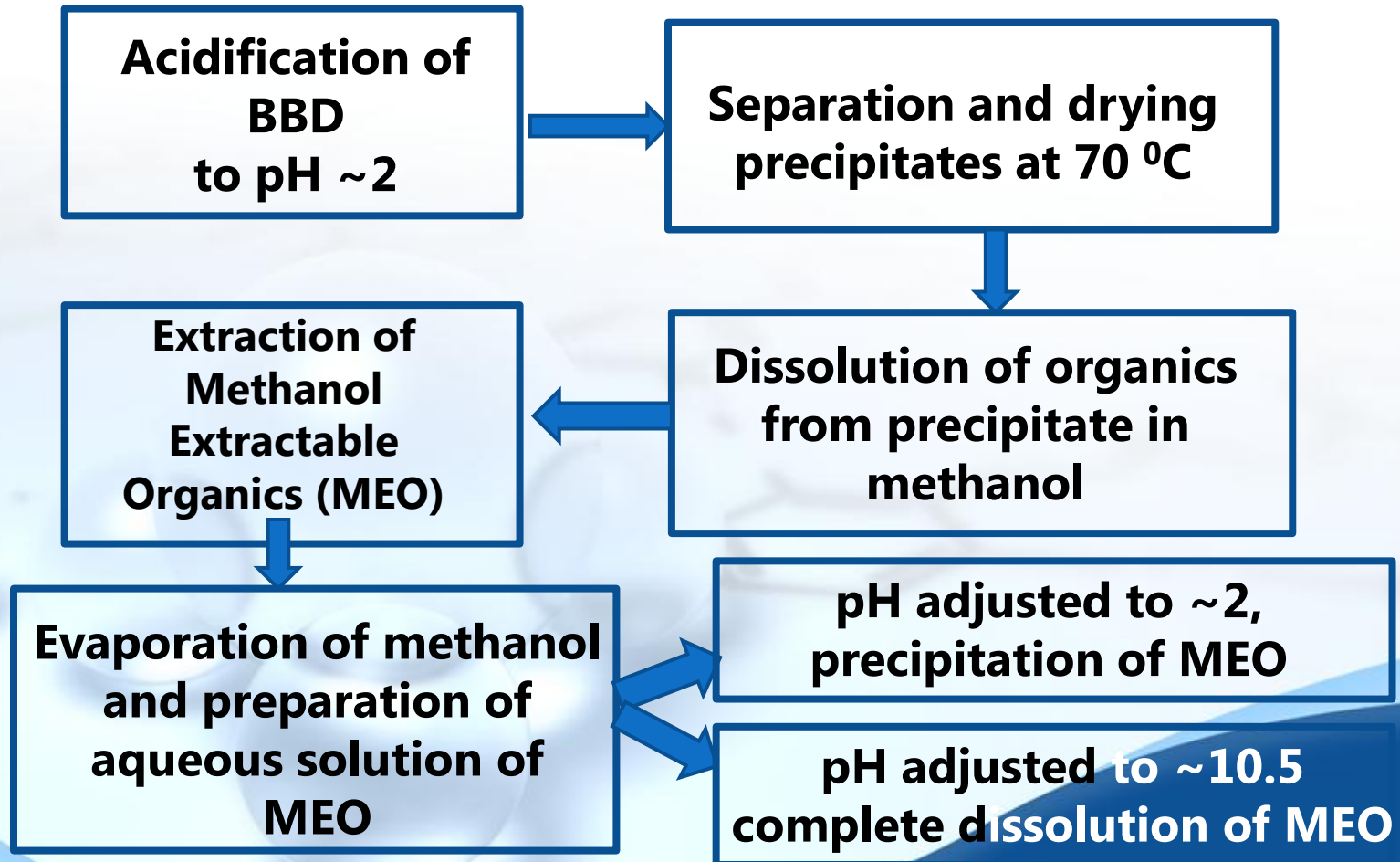
# Acidification of Blow-Down



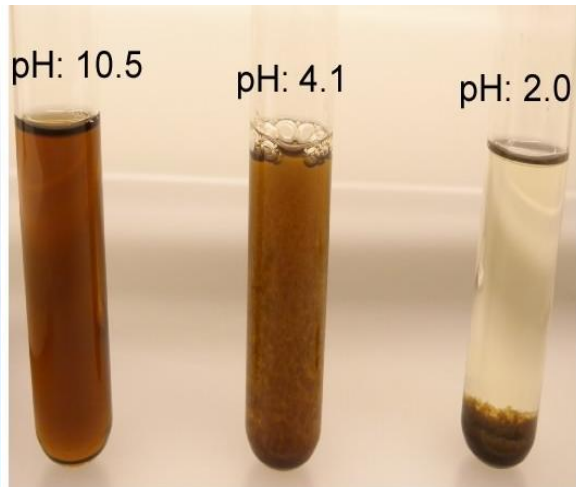
# Silica and DOM Removal on Acidification



# Methanol Extracted Organics (MEO)



# Effects of pH on Aqueous Methanol Extracted Organics

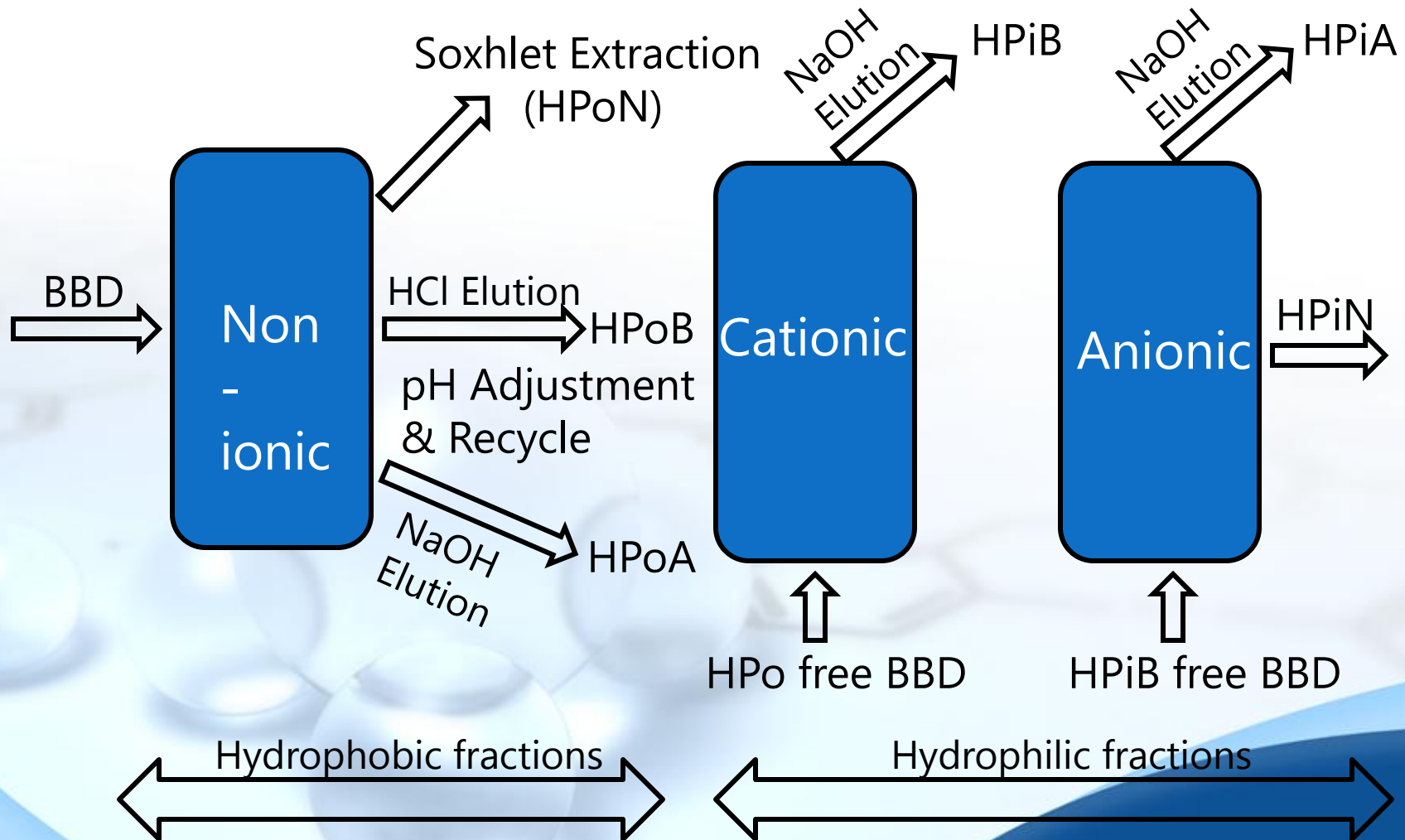


## CHNS analysis of acid precipitated organics

C (Wt %)	H (Wt %)	N (Wt %)	S (Wt %)
18.08	2.23	0.25	0.29

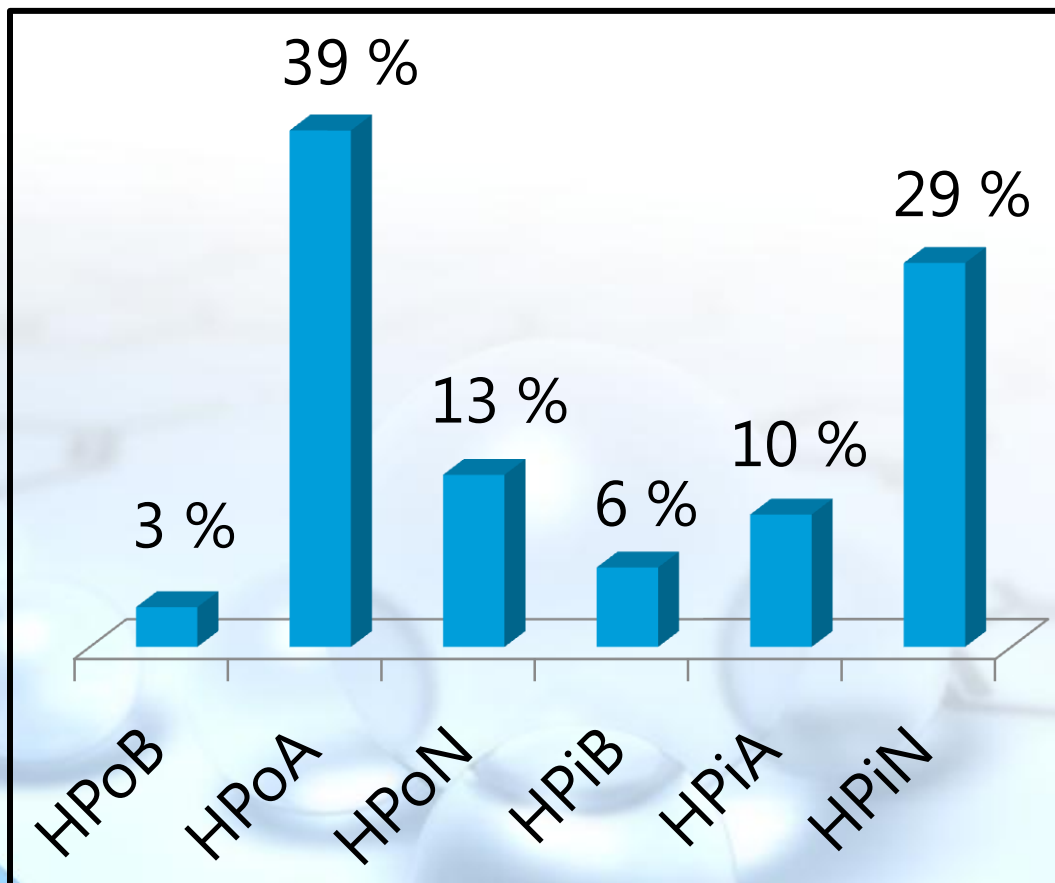
- The MEO fraction of DOM precipitates independently with acidification
- This behavior can be attributed to the protonation of organic acid salts to a free acid form, which become insoluble in aqueous media and precipitate at low pH

# Ion-Exchange Fractionation





# DOM Fractions in BBD



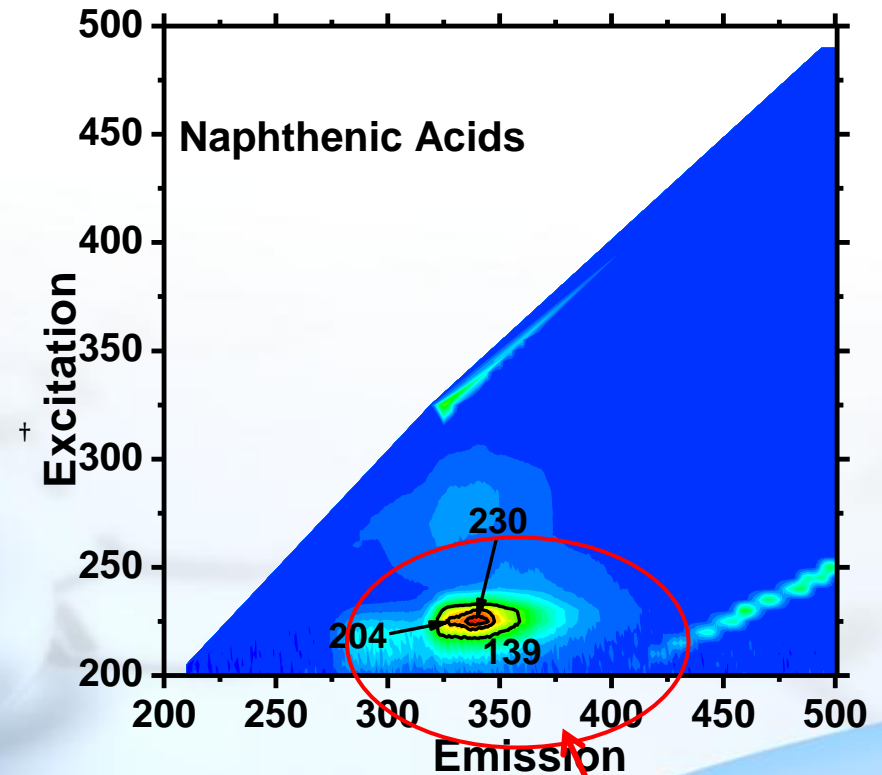
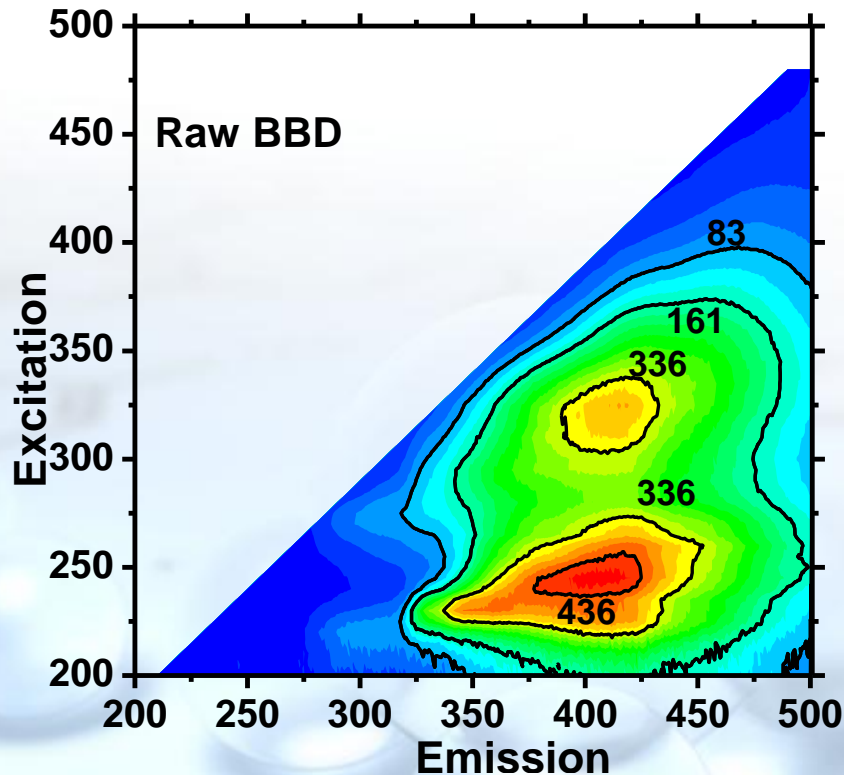
- Hydrophobic acid and hydrophilic neutral are the major fractions
- Hydrophobic neutral is the next major fraction

# Analytical Techniques

## Excitation-emission matrix (Fluorescence spectroscopy)

- Ex/Em ranges: 200 to 500 nm  
Step sizes: 10 nm for excitation  
5 nm for emission
- Other analytical techniques used are TOC analysis, SUVA<sub>254</sub>, ATR-FTIR spectroscopy

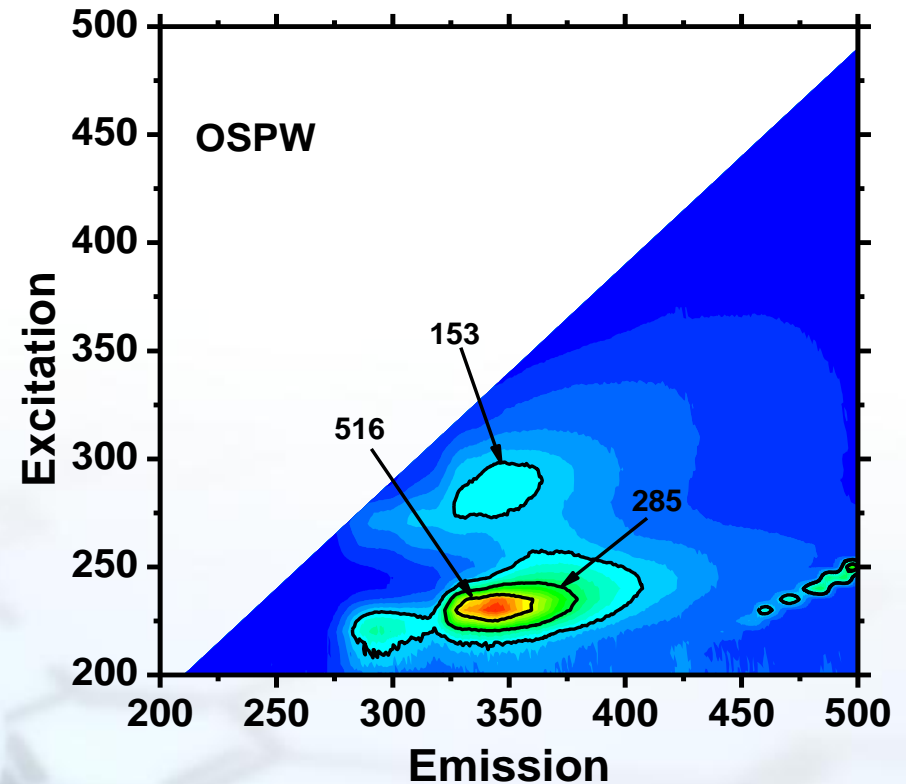
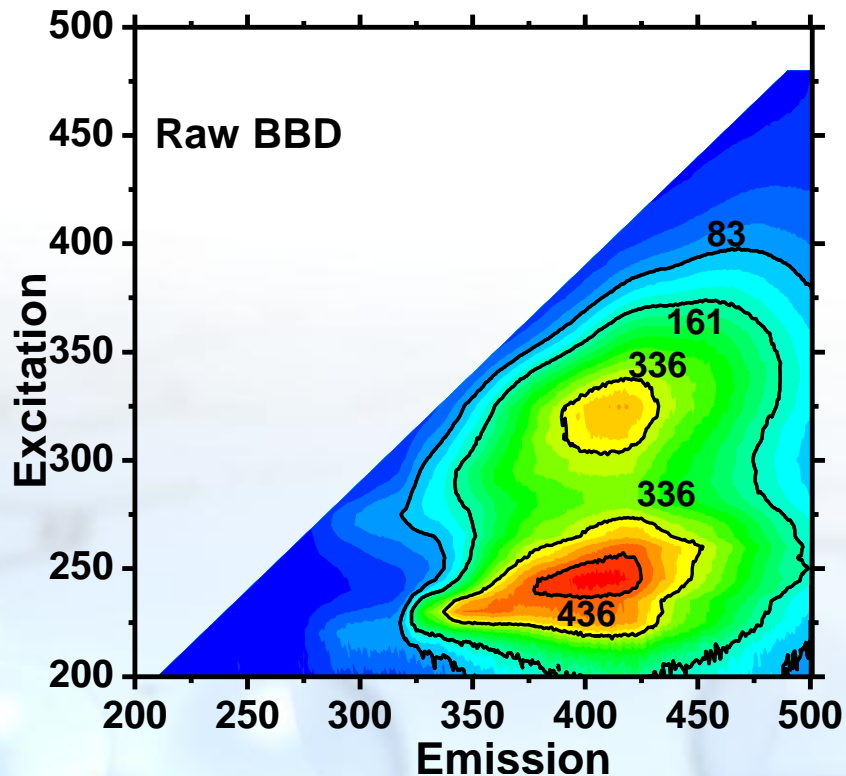
# EEMs of BBD- Organics and Naphthenic Acids



Signature of commercial Naphthenic acid (NA)  
not observed in BBD

Ex/Em 230/350

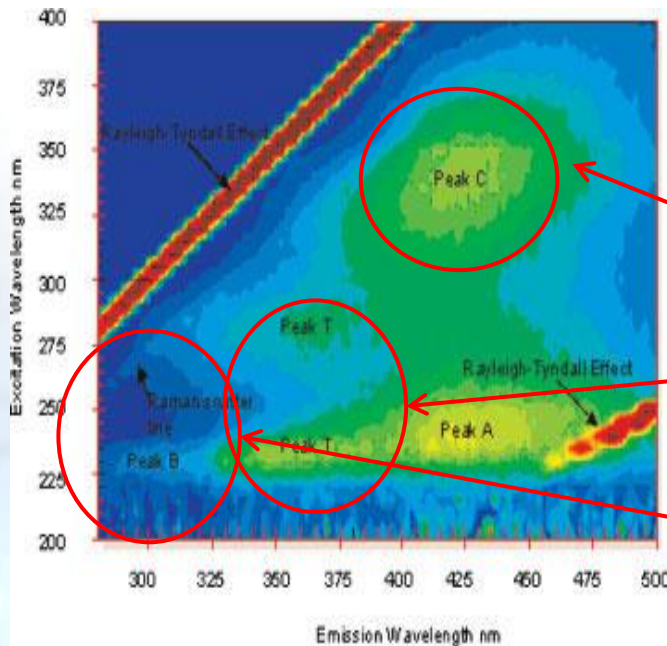
# BBD v/s OSPW



- The BBD does not contain NA signatures
- NA is a principal DOM component in OSPW

# **FLUORESCENCE SIGNATURES OF DOM FRACTIONS OBTAINED FROM ION-EXCHANGE FRACTIONATION**

# Typical NOM EEMs



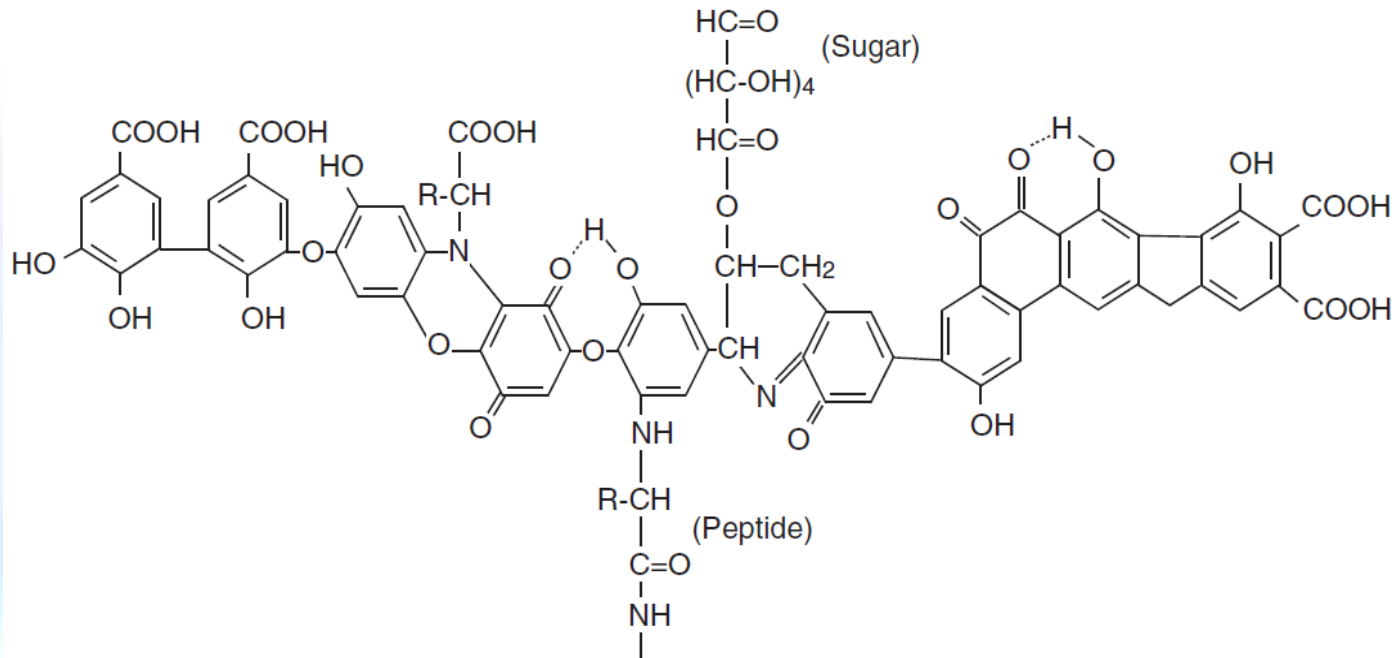
Fluorophore type	Excitation-emission wavelength (nm) fluorescence peaks
Humic like	<u>300-370/400-500</u>
Tryptophan	<u>225-237/340-381 and 275/340</u>
Tyrosine	<u>225-237/309-321 and 275/310</u>

**Hudson, N.; Baker, A.; Reynolds, D. Fluorescence analysis of dissolved organic matter in natural, waste and polluted waters-a review. River Research and Applications 2007, 23, 631-649.**



# Structures of Humic Acid

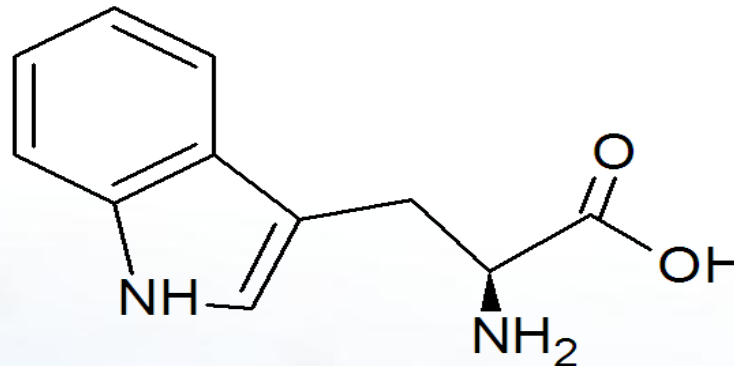
## Model Structure:



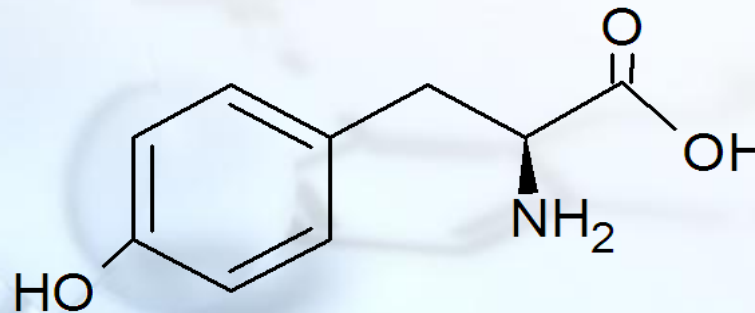
Aiken GR, McKnight D, Weshaw RL, MacCarthy P. 1985. An introduction to humic substances in soil, sediment and water. Humic Substances in Soil, Sediment and Water, John Wiley & Sons: New York; 203.

# Structures of Tryptophan and Tyrosine

Tryptophan

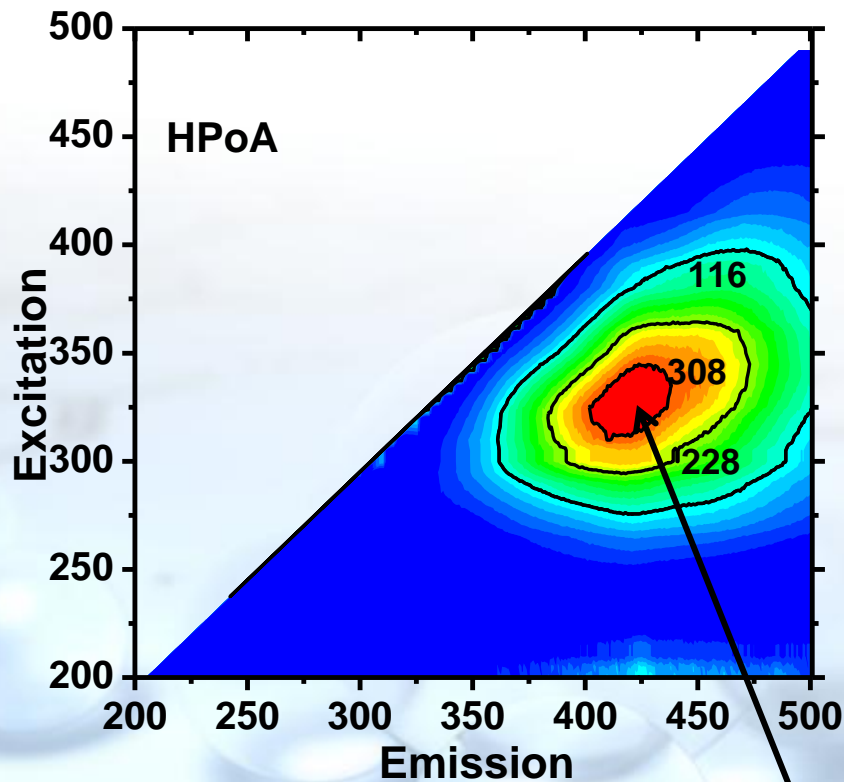


Tyrosine

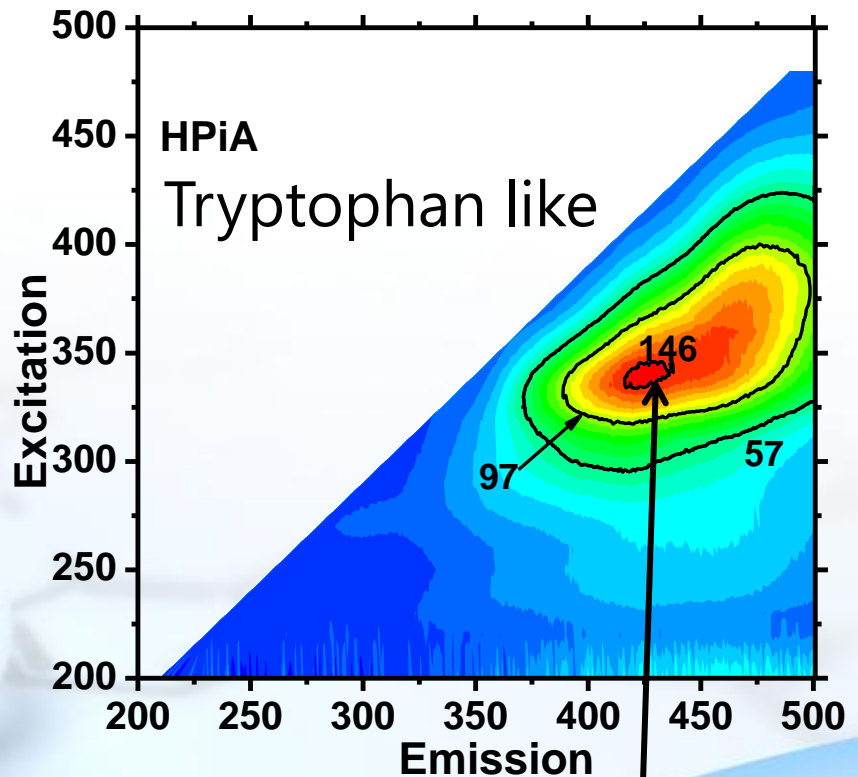


Hudson, N.; Baker, A.; Reynolds, D. Fluorescence analysis of dissolved organic matter in natural, waste and polluted waters-a review. *River Research and Applications* 2007, 23, 631–649.

# EEMs Contour of Acidic DOMs

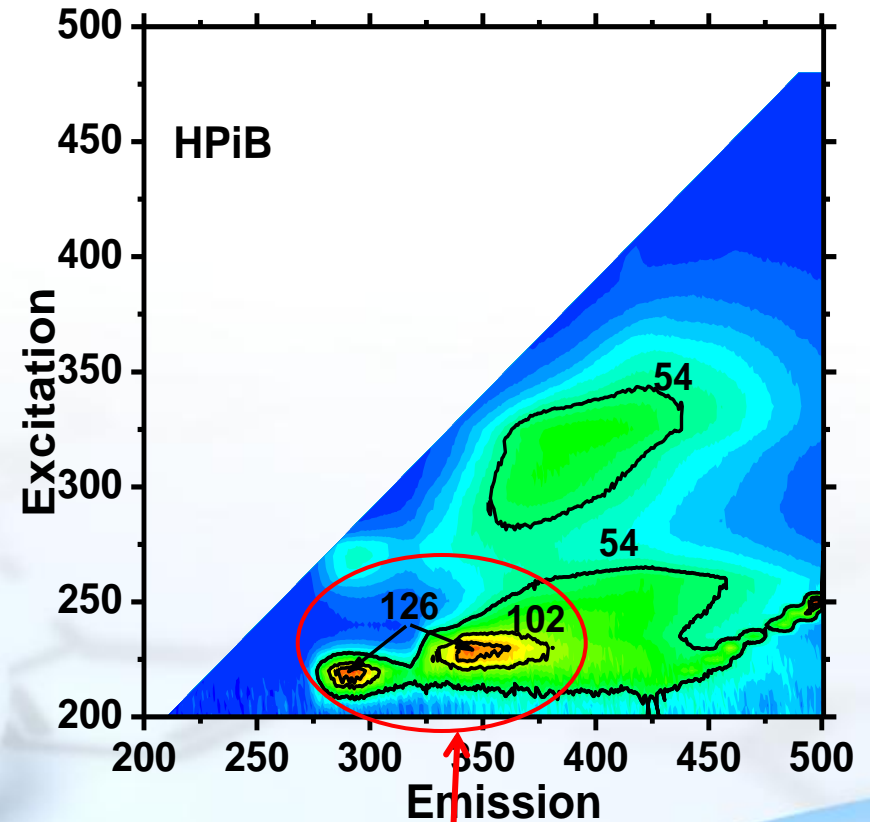
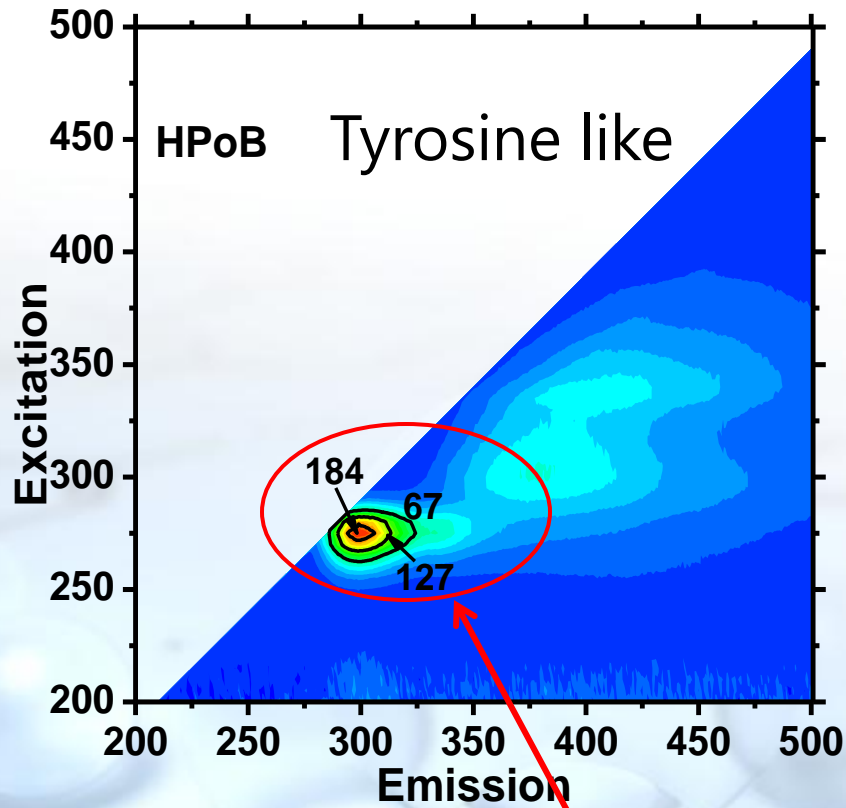


Humic like Ex/Em: 325/420

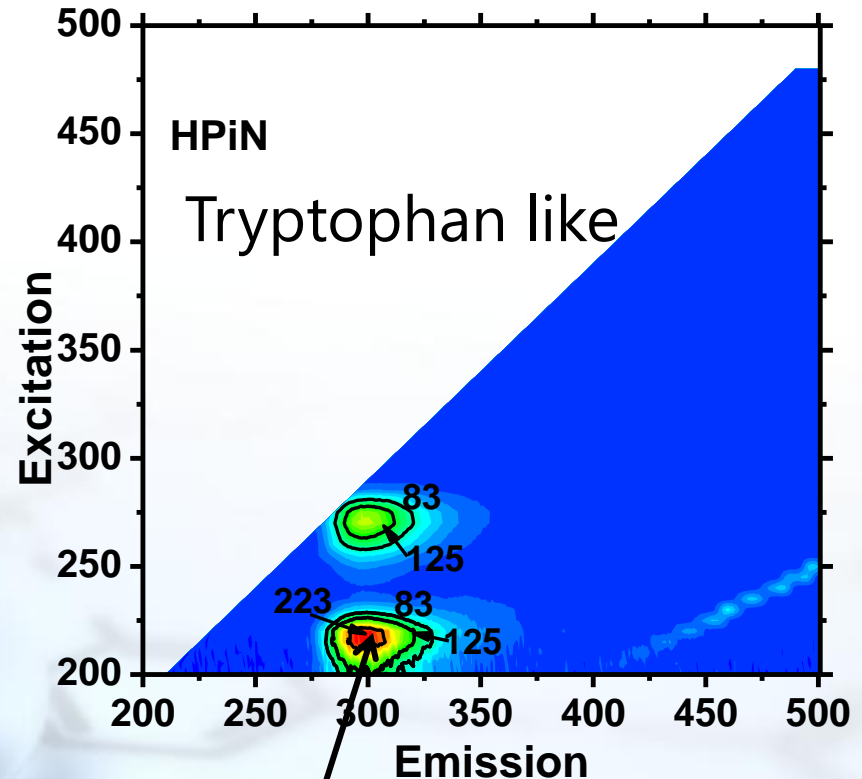
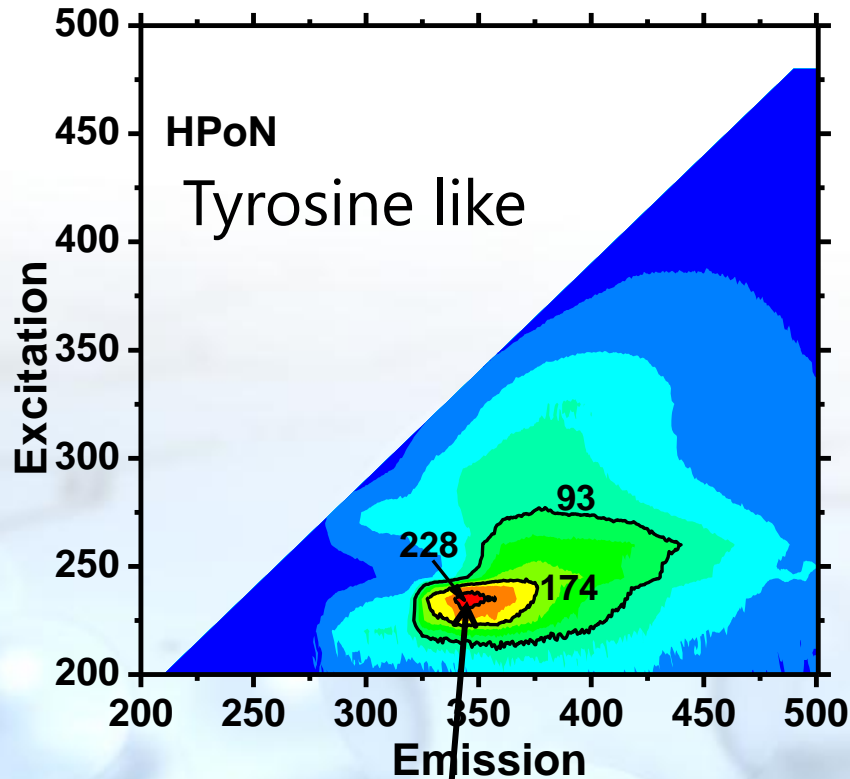


Ex/Em: 340/425

# EEMs Contour of Basic DOMs



# EEMs Contour of Neutral DOMs



# SUVA<sub>254</sub>

SUVA<sub>254</sub> value of DOM fractions:

Fraction s	SUVA <sub>254</sub>
HPoB	0.61
<b>HPoA</b>	<b>3.62</b>
<b>HPoN</b>	<b>3.95</b>
HPiB	1.11
HPiA	2.03
HPiN	0.18

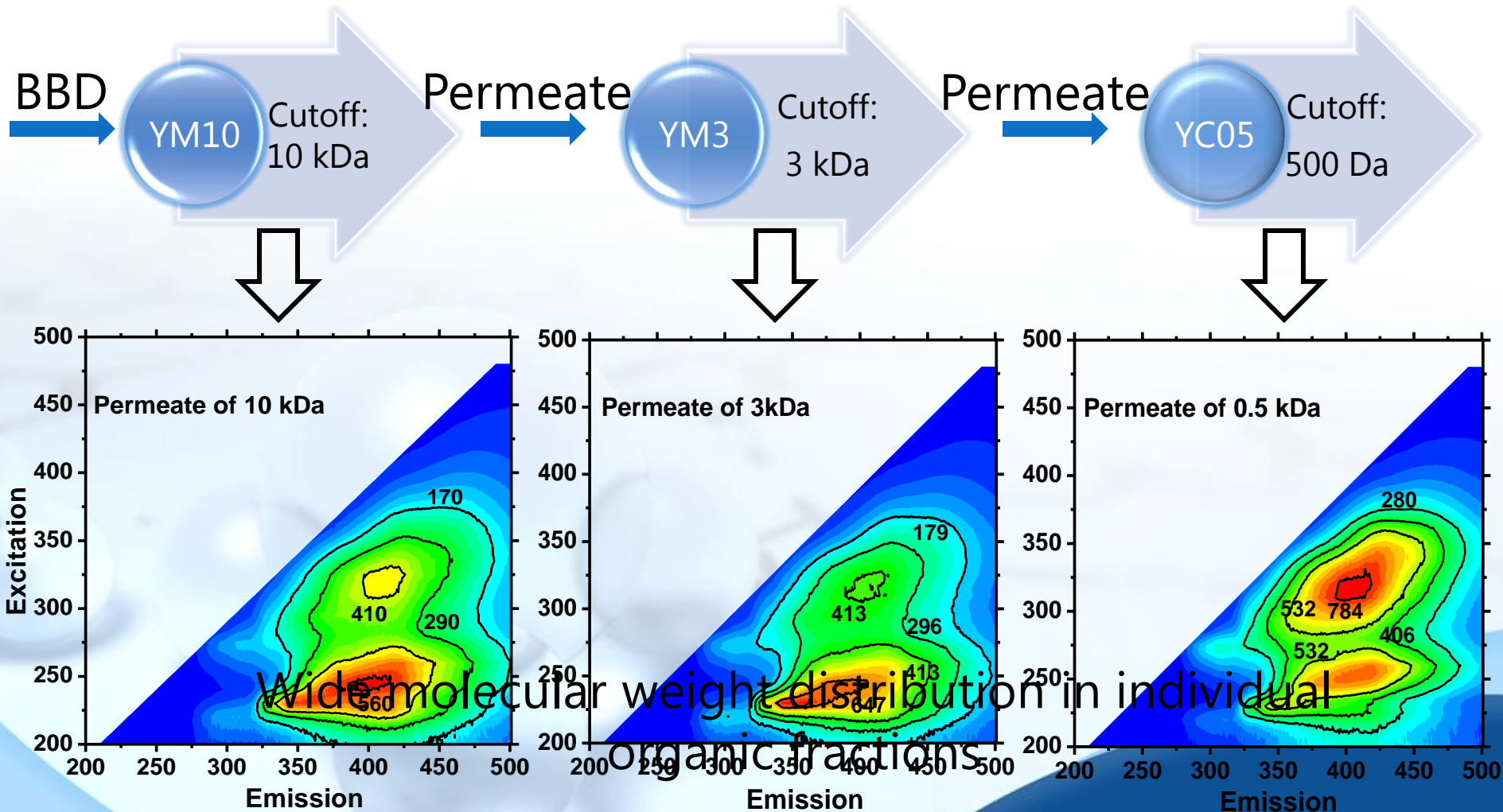
$$\text{SUVA}_{254} = (\text{UV Abs}_{254} / \text{TOC}) * 100$$

Presence of aromatic content in the HPoA and HPoN

Lower aromatic content in the other fractions



# Membrane Fractionation



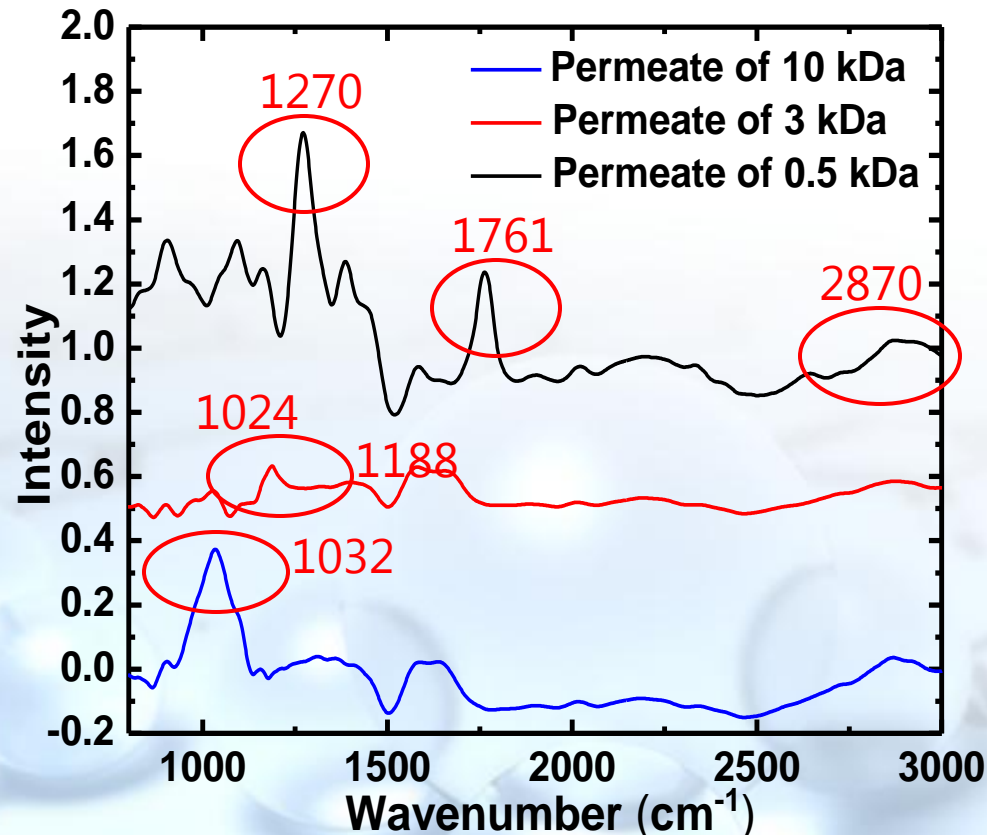
# Molecular Weight Distribution

Molecular weight range	Percentage DOM in permeate
> 10 kDa	8 %
> 3 kDa – 10 kDa	19 %
> 0.5 kDa – 3 kDa	33 %
< 0.5 kDa	40 %

Nanofiltration (500 Da) retains 60% of DOM; a significant fraction is too small for NF removal

40% of organic matter in BBD is of molecular weight less than 500 Da

# FTIR Results



10 kDa: Silica peaks at 1032  $\text{cm}^{-1}$

3 kDa: Very weak peaks obtained at 1024 and 1188  $\text{cm}^{-1}$

0.5 kDa: Peaks

- At 2870  $\text{cm}^{-1}$  C-H stretching
- At 1761  $\text{cm}^{-1}$  for C=O
- At 1270  $\text{cm}^{-1}$  for C-O

# Concluding remarks

- ◆ DOM in SAGD blow-down water consists of low molecular weight compounds
- ◆ Hydrophobic acids and hydrophilic neutrals are the major fractions in the SAGD blow-down water
- ◆ Membrane filtration is unable to separate the DOM fractions based on size exclusion
- ◆ Knowledge of physico-chemical characteristics of SAGD fluids is the first step toward building a program for treatment



# NSERC Industrial Research Chair in **Water Quality Management for Oil Sands Extraction**

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