



WorleyParsons

resources & energy

EcoNomics™

Different Approaches in Assessing Background Groundwater Quality

Ken Lyon, M.Sc., P.Geol., and Elizabeth Haack, Ph.D., P.Chem.

22 April 2010





Ken:

- ▶ What is “background”?
- ▶ Multiple lines of evidence approach

Elizabeth:

- ▶ Case studies
 - Aluminum
 - Barium

Both:

- ▶ Wrap-Up



Fallout from AENV Tier 1 Guidelines

- ▶ Alberta Environment (AENV) Tier 1 Guidelines:
- ▶ Several low guideline values from consideration of the Domestic Use Aquifer (DUA) and Freshwater Aquatic Life (FAL) receptors
- ▶ Groundwater quality “exceedances” often appear to be background, particularly metals
 - e.g. Al, As, Cd, Cu, Ba, Mn, Se, U, Zn
 - also nitrite & nitrate
 - chloride, sodium, chloroform (urban areas)



Site Impact or Background?

- ▶ Are Tier 1 exceedances impact from anthropogenic point sources or background?

 - ▶ What is background?
 - Natural background in absence of anthropogenic influence (can be used in place of Tier 1 guidelines)
 - Anthropogenic background from diffuse or non-point anthropogenic sources
- e.g. urban areas - road salting, leaky sewer & water lines, fertilizers



- ▶ AENV Guidance
 - Natural concentration in a particular groundwater zone in absence of any anthropogenic sources or activities
 - Will vary spatially & temporally (moving target)
 - Must consider groundwater quality data from several wells



► Site Challenges

- How many sites have several “background” wells?
- Who has baseline pre-development data and enough data for statistical or non-parametric analysis?
- Who has access to extensive databases in public domain?
- Who has budget for extensive analysis?
- What is “natural” in an urban or agricultural environment?



How best to find the answers? Multiple lines of evidence

- ▶ Five-point checklist of questions (may not have answers for all):
 - Origin (natural and/or anthropogenic)?
 - Compare with published data?
 - Correlation with other Site data?
 - Concentration distribution?
 - Controls on aqueous concentrations?



Where can parameters potentially originate from?

- ▶ Natural occurrences in the environment vs rural, urban, industrial, etc, anthropogenic occurrences

- ▶ Some sources of information:
 - CCME Environmental Quality Guidelines
 - Health Canada monographs
 - Groundwater geochemistry and pollution books
 - Other books, publications, reports
 - Experience



How do the data compare with published data or other reports?

- ▶ Some sources of information for Alberta sites:
 - Prairie Farm Rehabilitation Administration groundwater assessments
 - Farm well water quality studies
 - Geological Survey of Canada studies of the Paskapoo bedrock aquifer
 - Alberta Geological Survey reports
 - University research and theses
 - In-house consultant reports
 - Etc.



Are there apparent correlations with other site data or information?

- ▶ Apparent correlations
 - Locations and depths/ elevations?
 - Lithology and provenance?
 - Soil quality?
 - Sampling methods?
 - Groundwater quality and chemistry?

- ▶ Potential use of next-generation database management tools coupled with GIS software



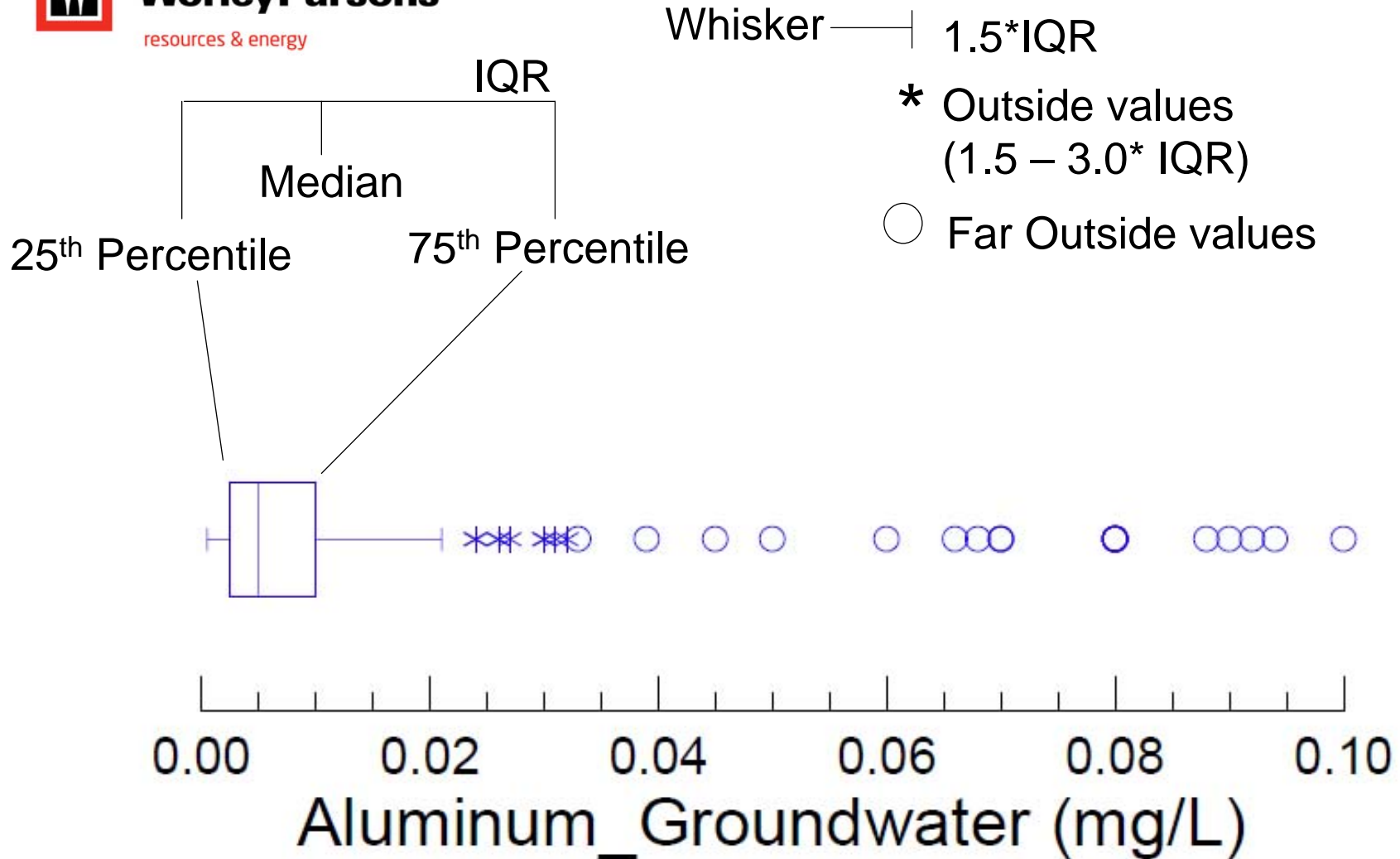
What does the distribution of data look like?

- ▶ Non-parametric box-and-whisker plots
- ▶ Five number summary:
 - median,
 - 25th & 75th quartiles
 - 1.5 & 3.0 x Interquartile Range
- ▶ 95% within 1.5xIQR if normally distributed



WorleyParsons

resources & energy



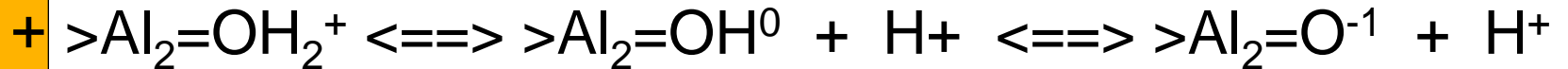


What controls concentrations in groundwater?

Mineral Surface

+ pH 

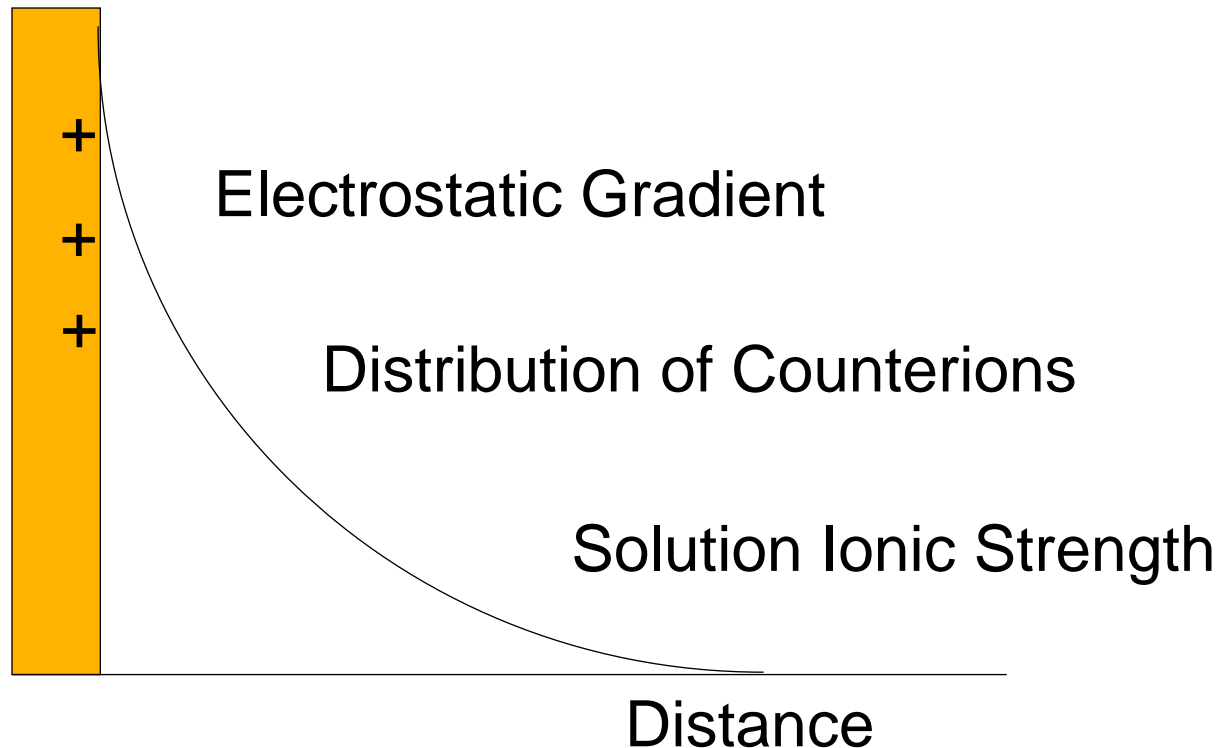
+ Variable Charge





What controls concentrations in groundwater?

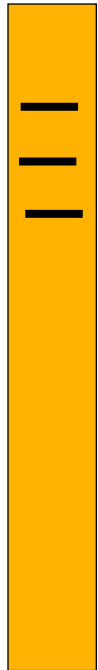
Mineral Surface





What controls concentrations in groundwater?

Mineral Surface

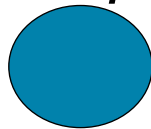


Permanent Charge

Cation Exchange Capacity (mmol charge)

$$\approx 2[\text{Ca}] + 2[\text{Mg}] + [\text{K}] + [\text{Na}] + 3[\text{Al}]$$

Species X



Speciation



What controls concentrations in groundwater?

- ▶ Cation or anion?
- ▶ pH?
- ▶ Redox or non-redox active?
- ▶ Expected speciation over site conditions?
- ▶ Solubility controls?
- ▶ Major or trace metal concentrations?
- ▶ Mineral or organic matter adsorption?
- ▶ Ionic strength of solution?
- ▶ Colloidal behaviour?



Aluminum Case Study (In-house Data for Calgary Area)

- ▶ Tier 1 guideline 0.1 mg/L (Receptor: FAL)
- ▶ Shallow Groundwater, Calgary, 300 analyses, several sites
- ▶ Occasional exceedances, ranging up to 1.7 mg/L

Multiple Lines of Evidence Checklist

- ▶ Potential Origins:
 - Natural sources: most abundant metal in earth's crust
 - General anthropogenic sources: vehicle parts, aircraft frames, pharmaceuticals, flocculants in water treatment (Health Canada 1998)
 - Potential site sources: Calgary drinking water 0.028 to 0.445 mg/L from water plant treatment (City of Calgary, 2009)



Aluminum Case Study continued

► Published Data:

- 13 of 816 Alberta farm well samples > 0.2 mg/L (Fitzgerald et al. 2001)

► Other Correlations:

- Most wells completed in silt, clay, Paskapoo mudstone indicating weathering of aluminosilicates
- Scattered locations

► Data Distribution:

- Upper box plot whisker (0.02 mg/L) about the same as upper limit for dissolved aluminum in near neutral pH range (Appelo & Postma 2007)

► Concentration Controls:

- Clay mineral gibbsite ($\text{Al}(\text{OH})_3$) crystals stable at diameter of 0.1 μm (Hem 1985)
- Compare field filtration 0.45 μm



- ▶ Conclusion: likely represent natural background



Barium Case Study

In-house Data Shallow Groundwater

- ▶ Tier 1 Guideline: 1 mg/L (Receptor: Domestic Use Aquifer)
- ▶ Rare to occasional exceedances, ranging up to 2-3 mg/L

Multiple Lines of Evidence Checklist

- ▶ Potential Origins (Health Canada 1990):
 - Natural sources: trace element igneous and sedimentary rocks, most commonly as barite (BaSO_4)
 - General anthropogenic sources: use in plastics, rubber, electronics, textiles, ceramic glazes and enamels, glass and paper, fuel oil additive, lubricant additives (drilling mud)
 - Potential site sources: fuel oil, drilling mud



Barium Case Study Continued

▶ Published Data:

- 2 of 816 Alberta farm well samples > 1 mg/L (Fitzgerald et al. 2001)

▶ Other Correlations

- No apparent lithologic correlation
- Scattered locations

▶ Data Distribution

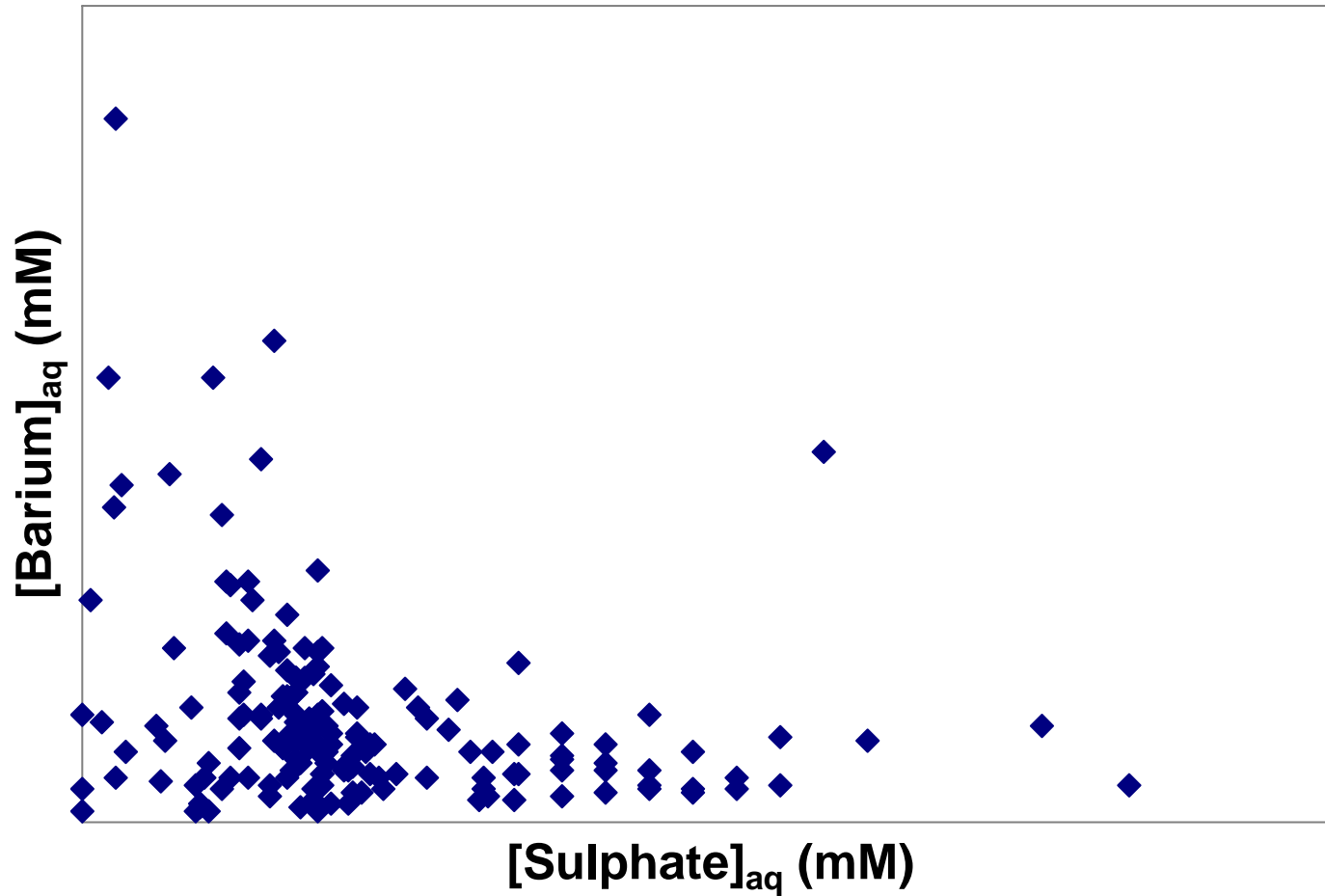
- Upper box plot whisker 0.7 mg/L

▶ Concentration Controls

- Solubility
- Sorption – Cation Exchange

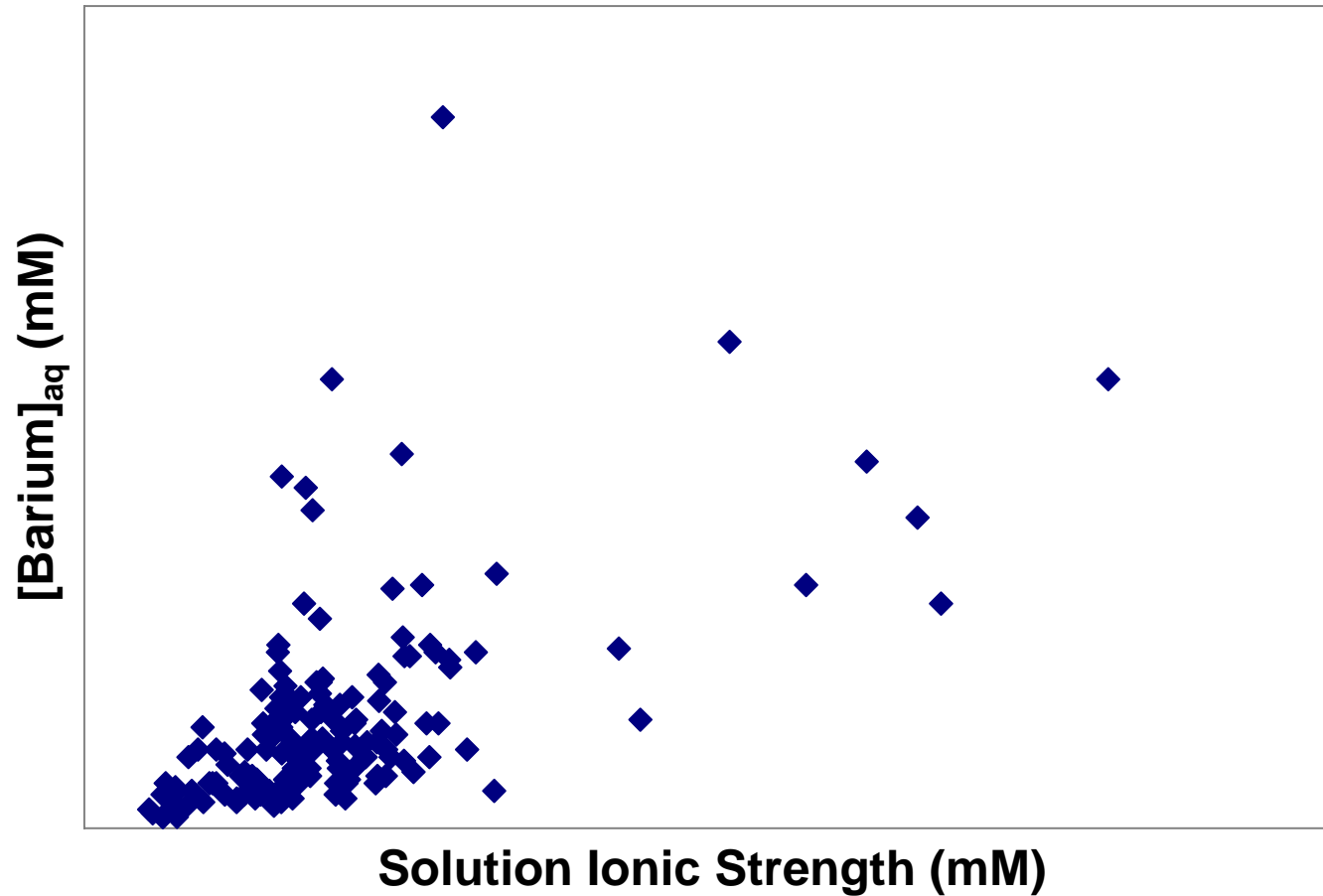


Barium Solubility Control Barium vs Sulphate





Barium Cation Exchange Control Barium vs Solution Ionic Strength





- ▶ **Conclusion: likely reflects anthropogenic source**
 - Direct from drilling muds or diesel fuel additives; or
 - Indirect from anthropogenic salinity related impacts



- ▶ Can be detailed but necessary exercise in dealing with Tier 1 Guidelines
- ▶ Becomes more efficient as knowledge “base” and database management systems developed
- ▶ Potential application to upstream well fields
- ▶ Ultimately an exercise in professional judgment
- ▶ Future challenges and study
 - Compilation and dissemination of data sets in the public domain



- Alberta Environment, 2009. Alberta Tier 1 Soil and Groundwater Remediation Guidelines. February 2009.
- Appelo, C.A.J. and Postma, D., 2007. Geochemistry, Groundwater and Pollution. A.A. Balkema Publishers, New York, NY. 649 p. (Corrected reprint 2007).
- The City of Calgary, 2009. Special Issue – 2008 Water Quality Report. Waterways, Vol. 11, Issue 3. Calgary, AB. July 2009.
- Fitzgerald, D., Chanasyk, D.S., Neilson, R.D., Kiely, D. and Audette, R., 2001. Farm Well Water Quality in Alberta. Water Quality Research Journal of Canada, Vol. 36, No. 3, pp.565-588.
- Grasby, S.E., Chen, Z., Hamblin, A.P., Wozniak, P.R.J., Sweet, A., 2008. Regional characterization of the Paskapoo bedrock aquifer system, southern Alberta. Canadian Journal of Earth Sciences. 45, pp.1501-1516.
- Health Canada, 1998. Chemical/Physical Parameters: Aluminum, Edited November 1998. Retrieved from <http://www.hc-sc.gc.ca/ewh-semt/pubs/water-eau/>. September 17, 2009.
- Hem, J.D., 1986. Study and Interpretation of the Chemical Characteristics of Natural Water. US Geological Survey Water Supply Paper 2254, Third Edition, Second Printing, Alexandria, Virginia.



WorleyParsons

resources & energy

THANK YOU!

ken.lyon@worleyparsons.com &
elizabeth.haack@worleyparsons.com